



EVALUATION OF PULP BROWNING IN MINIMALLY PROCESSED 'ROYAL GALA' APPLE TREATED WITH ERYTHORBIC ACID

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ABSTRACT

The production and marketing of minimally processed (MP) apple is limited by the rapid and intense browning of the pulp. The application of antioxidants is the main alternative to overcome this problem, but with limited results, mainly due to the lack of information about the most adequate concentration of antioxidant for a specific fruit under certain storage conditions, such as type of packaging, temperature, humidity and storage time. Therefore, the objective of this study was to evaluate the effect of different concentrations of sodium erythorbate on color, weight loss, total soluble solids (TSS), total acidity (TA), ratio (TSS / TA), total phenols, antioxidant activity, and enzymatic activity (peroxidase and polyphenoloxidase) in 'Royal Gala' apple MP during refrigerated storage. Each fruit was cut in four wedge shape pieces of similar size, and immediately immersed (1.0 min) in the antioxidant solutions: (a) distilled water - negative control; (b) L-cysteine chloride 0.6 % [m.v⁻¹] - positive control; (c) erythorbic acid 1.0 % [m.v⁻¹]; (d) erythorbic acid 2.0 % [m.v⁻¹] and (e) erythorbic acid 3.0 % [m.v⁻¹]. The activity of the enzymes peroxidase and polyphenoloxidase increased throughout the storage, however, in the apples where the antioxidants L-cysteine (0.6 %) and erythorbic acid (1%, 2% and 3%) were applied the enzymatic activity was lower than the control. The 3 % erythorbic acid, in addition to satisfactorily preserving the color, preserved the phenolic compounds and the antioxidant activity during the nine days of refrigerated storage.

1. Introduction

The demand for minimally processed (MP) fruits and vegetables is growing in the food market. However, consumers look for fresh products with organoleptic characteristics close to *in natura* equivalents (Tappi et al., 2017). The preparation of MP products consists of a series of steps, such as sanitizing, cutting, packaging, storage, marketing, among others (Segura-Ponce et

al., 2018). These steps decrease the products shelf life (Putnik et al., 2017) and cause enzymatic browning (EB), especially in apples (*Malus domestica* Borkh). Thus, the control of EB is of great importance, as it occurs in many fruits and vegetables and negatively affect attributes of color, taste and nutritional value (Holderbaum et al., 2010, Ioannou and Ghoul 2013). Dogan and Dogan (2004) observed that colour changes

in apple pulp occur due to the high concentration of phenolic compounds in this fruit. This occurs because cutting and other mechanical damage allow oxygen penetration into tissues, resulting in rapid darkening by the oxidation of phenolic compounds. In this process, quinones are polymerized with other quinones or with phenolic compounds, resulting in the appearance of brown pigments (Kim et al., 2017). The phenolic substrates involved in the darkening reaction (e.g. 5-caffeoylquinic acid) are separated from polyphenoloxidase (PPO) enzyme inside intact organelles (vacuoles and plastids, respectively), thus inhibiting the darkening reaction. The cutting, or other mechanical damage during minimal processing, allow enzymes and substrates to react, causing tissue browning. Genetically modified apple cultivars, resistant to enzymatic darkening, have already been successfully obtained (Espley et al., 2013). However, although it is already in the commercialization phase, these cultivars remain little diffused due to their compatibility with different edaphoclimatic conditions and also the consumer resistance to consuming genetically modified products. In this context, one of the alternatives to inhibit enzymatic darkening in MP apples is the use of antioxidant agents. A commonly used agent is L-cysteine; this amino acid contains a thiol group with reduction properties (Richard-Forget et al., 1992). Several studies have demonstrated the efficiency of L-cysteine in the inhibition of pulp darkening of MP products such as 'Red Delicious' apples (Eissa et al., 2006), Fuji apples (Moreno et al., 2016), 'Stylist' lettuce (Bernardo et al., 2015) and Lychia (Ali et al., 2016). Similarly, the antioxidant erythorbic acid (D-isoascorbic acid) (Kall and Andersen 1999, Sun et al., 2013) is a

stereoisomer of ascorbic acid but without the activity of vitamin C. This antioxidant has the same properties of its stereoisomer but five times cheaper (Pineli et al., 2005, Martin-Belloso and Fortuny 2010). The use of EA as an antibrowning agent was evaluated in MP 'Ágata' potatoes (Pineli et al., 2005), canned apple and beer (Andersen 1999). In addition, EA is considered a safe food antioxidant in the European Union and the US, when used according to its legislation (Sun et al., 2013, EFSA 2016). Despite the proven antioxidant properties, there are few reports in the literature evaluating the use of EA as a darkening inhibitor in MP apples.

In this context, the objective of this study was to test different concentrations of EA for the maintenance of physicochemical attributes in MP 'Royal Gala' apples stored in refrigerated environment for up to nine days.

2. Materials and methods

2.1. Apples and maturity indices

Samples of apple (*Malus domestica* Borkh 'Royal Gala') were harvested in the year of 2014, from a commercial orchard located in the municipality of Vacaria, Rio Grande do Sul, Brazil (28° 30' 44" S, 50° 56' 02" O). Apples were harvested considering their commercial mature stage, based on the average starch content (4.80) measured according to Travers et al., (2002), on a scale from 1 (one) to 10 (ten), where 1 and 10 correspond to the maximum and minimum content of starch, respectively; pulp firmness (12.84 N); concentration of total soluble solids (12.03 °Brix) and titratable acidity (0.22 g of malic acid 100 g⁻¹ FW), measured according to the methods described below. Samples were selected according to size, the absence of visible mechanical damage and rot. Fruits were temporarily stored (few weeks) at

1.0 °C, relative humidity of 90.0 % ± 5.0 %, and finally used in the present study at the Postharvest Physiology Laboratory, Food Center, Embrapa Clima Temperado.

2.2. Sanitization, cutting and dipping

Fruits were washed and sanitized by dipping into a sodium hypochlorite solution (100 ppm, pH 6.5, at 6.5 °C ± 1.5 °C), for 10 min. Each fruit was cut in four wedge shape pieces of similar size; the central core and seeds were discarded and the fruit epidermis was preserved. After cutting, the apple pieces were dipped for 1.0 min in the following liquids: **(a)** distilled water - control (CT), **(b)** L-cysteine chloride 0.6 % [m.v⁻¹] positive control (LC), **(c)** erythorbic acid 1.0 % [m.v⁻¹] (EA 1.0 %), **(d)** erythorbic acid 2.0 % [m.v⁻¹] (EA 2.0 %) and **(e)** erythorbic acid 3.0 % [m.v⁻¹] (EA 3.0 %). After the immersion, the apple pieces were drained for 5 min, placed in trays of expanded polystyrene, wrapped with PVC film (9 µm thick) and stored for different periods (0 d, 3 d, 6 d, 9 d) at 4.0 °C ± 1.0 °C and relative humidity of 90.0 % ± 5.0 %, for simulation of shelf-life.

2.3. Physicochemical analysis

2.3.1. Color measurement: measured on the equatorial region of the wedge-shape pieces of apple, on the pulp, using a Minolta CR-400 colorimeter with a CIE L*a*b* reading system, proposed by the *Commission Internationale de l'Eclairage* (CIE). These parameters were used to calculate the browning index (BI) according Palou et al., (1999).

2.3.2. Mass loss: measured according to Pereira et al. (2006) with the formula $ML (\%) = (M_i - M_f) / M_i \times 100$, where M_i and M_f correspond to the apple pieces initial mass and final mass, respectively;

2.3.3. Pulp firmness: measured according to Melo et al., (2009) using a Texture Analyser (TA XT plus 40855, Stable Microsystems, England) with a 2 mm diameter probe, penetration depth of 5 mm, pre-test velocity of 1.0 mm s⁻¹; 2.0 mm s⁻¹ test; post-test of 10.0 mm s⁻¹ and force of 5 kg. The readings were performed in the middle portion of the pieces and the results were expressed in Newton (N).

2.3.4. Soluble solids: determined by using an Atago refractometer (ATAGO, model PAL⁻¹), with results expressed as °Brix.

2.3.5. Titratable acidity: obtained by titration of 0.1 M NaOH solution into 100 mL sample solution (10 mL mashed pulp + 90 mL distilled H₂O) using a digital burette (Brand[®]) until pH stabilization at 8.1; the results were expressed in mg of malic acid equivalents (MAE) per 100g of pulp (FW).

2.3.6. Antioxidant activity (DPPH): evaluated using the method described by Brand-Williams et al., (2005), with some modifications. First, 10 mL of methanol was added to 2.5 g of fresh apple and homogenized for 1.0 min (ultra-turrax homogenizer, IKA). Extracts were centrifuged at 4000 rpm, 30 min, 1.0 °C (Eppendorf – Centrifuge 5810R - Rotor F-45-30-11). The supernatant was collected and stored at -80 °C until analysis. Apple extract (100 µL) was added to 3900 µL DPPH[•] solution (in methanol), and the reaction mixture was kept in the dark for 24 h. After this period, the absorbance was spectrophotometrically read at 515 nm. The results are expressed as mg of Trolox equivalent per 100 g of FW.

2.3.7. Total phenolic compounds: measured according to the Folin–Ciocalteu method adapted from Swain and Hillis (1959).

Briefly, 250 μL aliquot of the extracts (the same used for DPPH $^{\cdot}$ analysis) was combined with 250 μL of 0.25 M Folin-Ciocalteu reagent and 4000 μL ultrapure water. After 3 min of reaction, 500 μL of 0.5 M Na_2CO_3 was added, following incubation for 2 h at room temperature and absorbance reading at 725 nm. The results were expressed as grams of chlorogenic acid equivalents (CAE) per 100 g of FW. A chlorogenic acid standard curve (0.0 mg mL^{-1} to 0.5 mg mL^{-1}) was used.

2.3.8. Polyphenoloxidase enzyme (PPO) activity: determined according to the adapted methodology described by Cano et al., (1997), based on the increase of the absorbance (420 nm) rate at 25 $^{\circ}\text{C}$. Apple extracts were obtained as follows: 10 mL of phosphate buffer (0.2 M, pH 7), containing 0.2 g of polyvinylpyrrolidone (PVP), was added to 5 g of apple and homogenized for 1 min (ultra-turrax homogenizer, IKA), filtered and centrifuged at 16000 g. An aliquot of 0.1 mL of supernatant was then mixed with 2.9 mL of catechol (0.11 M in phosphate buffer 0.5 M, pH 7.0). The enzymatic activity was spectrophotometrically monitored at 420 nm for 3 minutes. The results are expressed as $\Delta\text{A}_{420} \text{ min}^{-1}\text{g}^{-1}$.

2.3.9. Peroxidase enzyme (POD) activity: determined according to the adapted methodology described by Cano et al., (1997), based on the increase of absorbance (485 nm) rate, at 25 $^{\circ}\text{C}$, using the same extract obtained for polyphenoloxidase analysis. An aliquot of 50 μL of apple extract was mixed with 2.7 mL of phosphate buffer (0.05 M, pH 7.0), 0.1 μL of hydrogen peroxide (1.5 %, v v $^{-1}$) and 200 μL of guaiacol. The enzymatic activity was calculated based on the increase of absorbance at 485 nm as a function of time

(3 min). The results are expressed as $\Delta\text{A}_{485} \text{ min}^{-1}\text{g}^{-1}$.

2.3.10. Statistical analysis: The experimental design was completely randomized in a factorial scheme with three biological replicates and three analytical replicates. Treatment factors were the storage time (0 d, 3 d, 6 d and 9 d) and the liquids for dipping the apple pieces: distilled water; 0.6% L-cysteine and erythorbic acid (1 %, 2 % and 3 %). Data normality was determined by Shapiro-Wilk test and variance homoscedasticity by the Hartley test. Later, the results were submitted to analysis of variance (One-Way ANOVA) ($p < 0.05$); when significant, the means were compared by the LSD test ($p < 0.05$).

3. Results and discussions

Enzymatic browning is the main cause of decline in quality and shelf life of minimally processed apples. Therefore, when evaluating the color parameters, there were significant ($p < 0.05$) differences in the variables L*, a*, b* and darkening index. In relation to the L* coordinate (Figure 1a), all treatments with antioxidants, LC and EA (1 %, 2 % and 3 %), remained stable and higher than CT during storage. After 9 d, EA 1 % presented a small decrease and CT a small increase in L* value, but without statistical differences. These results were different from those obtained by Vilas-Boas et al., (2015) when applying antioxidants to MP 'Williams' pears. Regardless of the antioxidant used, these authors observed a decrease in the L* value along the refrigerated storage. The inactivation of POD and PPO enzymes by erythorbic acid acidification of pulp surface may have contributed to the maintenance of luminosity during storage. In addition, Qi et al., (2011) reported that storage at low temperature contributes to delay the loss of luminosity in MP apples.

Regarding a^* coordinate (Figure 1b), LC and EA (1 %, 2 % and 3 %) also provided a satisfactory control in the darkening of the pulp until the third day of storage, since negative values of a^* indicate pulp with a greenish-yellow coloration. From the sixth day on, only EA 3 % treatment was capable of maintaining the same coloration observed on the first day (0 d) of storage. In contrast, apples with reddish flesh (positive values of a^*) were observed under the treatments LC and EA

(1 % and 2 %), until the sixth day of storage. The reddish tint observed in apples treated with LC is possibly due to the regeneration of phenolic compounds; this occurs when low concentrations of this antioxidant are used (Martins et al., 2015). However, the satisfactory effect observed with EA 3 % can be attributed to its strong reducing properties (Carocho et al., 2018), since this compound can react with oxygen and thus remove it from a closed system (Lee et al., 2012), avoiding the darkening of the pulp.

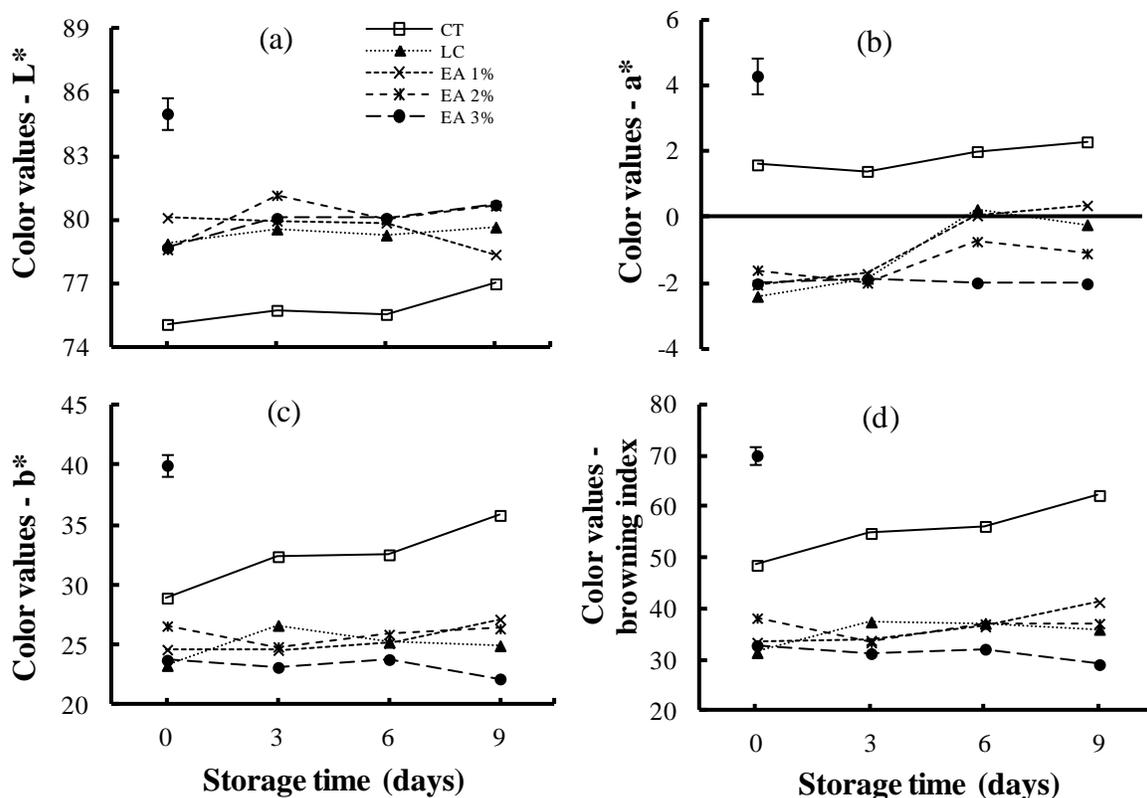


Figure 1. Color values, L^* (*LSD = 0.45), a^* (*LSD = 0.27), b^* (*LSD = 0.78) and browning index (*LSD = 1.67) measured in minimally processed 'Royal Gala' apples stored for 0 d, 3 d, 6 d, and 9 d at $4.0 \text{ }^\circ\text{C} \pm 1.0 \text{ }^\circ\text{C}$, relative humidity of $90.0 \% \pm 5.0 \%$, after treatment with distilled water (CT) as negative control; L-cysteine chloride (LC) 0.6 % (m.v^{-1}) as a positive control; erythorbic acid (EA): EA 1 % (m.v^{-1}), EA 2% (m.v^{-1}) and EA 3% (m.v^{-1}). *Vertical bars indicate the Least Significant Difference (LSD) at $p \leq 0.05$.

Regarding the b^* coordinate (Figure 1.c), although all antioxidants prevented pulp darkening, the EA 3 % treatment

resulted in a lighter pulp throughout the storage period. The process of oxidative staining is triggered by the membrane

rupture inside cells (Toivonen 2004), which results in the mixture of substrates (polyphenols) with PPO enzyme (Cortellino et al., 2015). In the presence of oxygen, PPO catalyse the monophenols hydroxylation into diphenols and subsequently the oxidation of diphenols in quinones (Cortellino et al., 2015). The former reaction is relatively slow and results in colorless products; the latter is relatively rapid and results in colored quinones. Subsequent reactions that occur after quinones production lead to the melanin accumulation, a brown pigment that gives the oxidized MP products their characteristic color (Cortellino et al., 2015). Hence, b^* values indicate a satisfactory action of the tested antioxidants, with emphasis on the EA 3 %. This result is similar to that observed by Pizato et al., (2013) when working with protein isolates on 'Gala' apples.

As for the browning index (Figure 1d), the different EA concentrations (1 %, 2 % and 3 %) and LC maintained pulp darkening lower than CT during the evaluation period. The EA 3 % kept the color of MP apples very close to that observed on the first day of storage (Figure 2), being the best treatment. It should be noted that the more intense yellow observed near the seed cavity (endocarp), in apples treated with EA 3%, does not indicate ineffectiveness of this antioxidant, since this region is naturally more yellowish than the fleshy pulp (mesocarp). The satisfactory result obtained with EA 3% is important because according to Altisent et al., (2014), the maintenance of MP fruit color indicates freshness and quality. The darkening inhibition process triggered by L-cysteine in apples pulp occurs through conjugation with *o*-quinones to form colorless compounds; another mechanism is the reduction of *o*-quinones to

their phenolic compound precursors (Koblitz 2000). According to Richard-Forget et al., 1992, the compounds resulting from cysteine and *o*-quinones conjugation may act as competitive inhibitors of PPO, being another mechanism of darkening inhibition. However, when there is a quinone excess and all of the L-cysteine has been consumed, the former can react with the cysteine-quinone addition compounds, giving rise to violet pigments. This explains the slightly browning observed at the end of the storage in apples treated with LC 0.6 %. In addition, LC contains sulphur in its constitution, giving rise to volatile sulphur compounds during its metabolization; this alters the taste and characteristic odour of the products treated with this antioxidant. On the other hand, the EA provides color maintenance; without compromising the taste, aroma and nutritional quality in MP products (Salata et al., 2014). Because of that, there are advantages in its use in comparison to LC, in addition to its low cost. According to Mishra et al., (2012), the EA prevents the darkening process by inhibiting the PPO enzyme and reducing quinone intermediates back to the diphenols (Mishra et al., 2012). Rojas-Grau et al., (2008), when working with ascorbic acid (AA), an ester of erythorbic acid, reported that although several authors have demonstrated the efficiency of AA in the control of enzymatic darkening, its efficacy is lower than that of L-cysteine. According to the authors, when AA is completely oxidized to dehydroascorbic acid, the quinones may again accumulate and cause browning. In view of this, it is assumed that the good results obtained with EA 3 % are due to the higher concentration of this salt, retarding its oxidation and thus preventing pulp and bundle sheath darkening in MP apples.

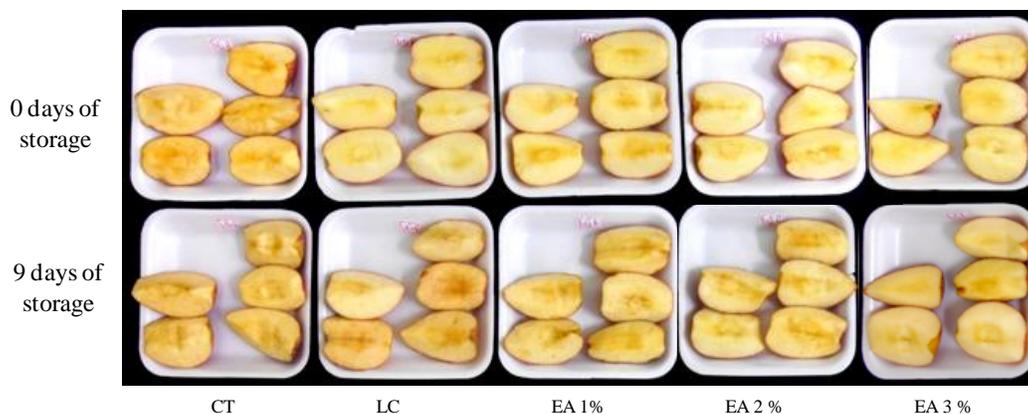


Figure 2. Pulp darkening of minimally processed apples, treated with distilled water, as negative control treatment (CT); L-cysteine chloride (LC) 0.6 % (m.v⁻¹), as a positive control; erythorbic acid (EA): EA 1 % (m.v⁻¹), EA 2 % (m.v⁻¹) and EA 3 % (m.v⁻¹); stored for 0 d, 3 d, 6 d, and 9 d at 4.0 °C ± 1.0 °C, and relative humidity of 90.0 % ± 5.0 %.

The *ratio* (total soluble solids / titratable acidity) is commonly used to evaluate fruit quality. A change in the *ratio* might affect how the apple taste (Piagentini and Pirovani 2017). This relationship (Figure 3a) had a small decrease in all treatments until the sixth day of storage, with a subsequent increase until the ninth day of storage. These changes can be explained by the metabolism

of organic acids in the respiratory pathways and subsequent conversion into non-acidic molecules (Pech et al. 2008). Total soluble solids can also accumulate by protopectin hydrolysis into soluble pectin or by starch hydrolysis into glucose and fructose (Barnes and Anderson 2018; Liu et al. 2018).

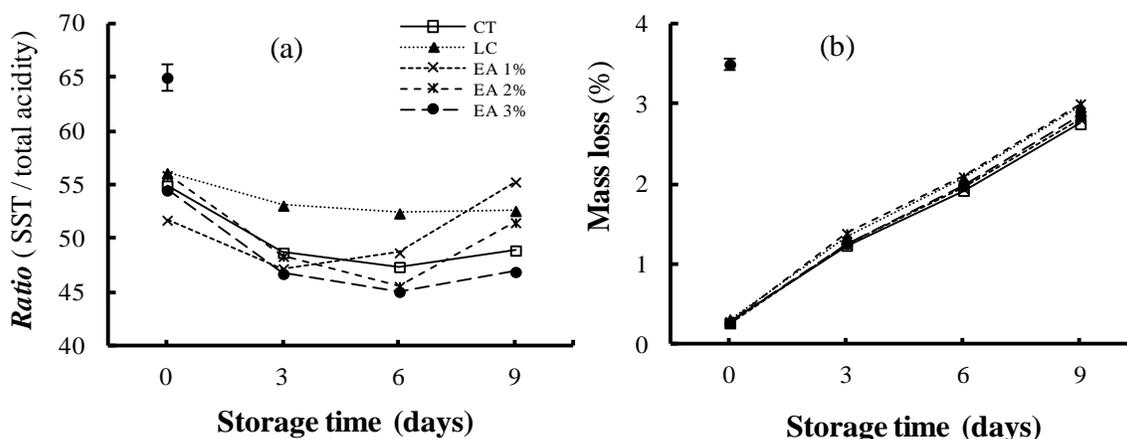


Figure 3. *Ratio* (*LSD = 1.21) and mass loss (*LSD = 0.067); measured in minimally processed 'Royal Gala' apples stored for 0 d, 3 d, 6 d, and 9 d at 4.0 °C ± 1.0 °C, relative humidity of 90.0 % ± 5.0 %, after treatment with distilled water as a negative control treatment (CT); L-cysteine chloride (LC) 0.6 % (m.v⁻¹) as a positive control; and erythorbic acid (EA): EA 1 % (m.v⁻¹), EA 2 % (m.v⁻¹) and EA 3 % (m.v⁻¹). *Vertical bars indicate the Least Significant Difference (LSD) at p ≤ 0.05.

The mass loss, a limiting factor in the shelf life of MP fruits and vegetables, increased during storage; with a mean mass loss (Figure 3b) of 2.6 % at the end of the storage period. This loss of mass was 5 times lower than that observed by Sumonsiri (2017) when working with ascorbic acid and nisin on 'Fuji' MP apples. The gradual mass loss throughout storage is due to pulp exposure to the atmosphere, potentiating the loss of water (Pajak et al., 2017). Despite the significant differences observed between treatments, the mass loss was minimal, not compromising the final quality of the product. The pulp firmness (data not shown) did not differ between treatments and along the storage period, with a mean value of 2.5 N.

Apples contain large amounts of endogenous phenolic compounds with antioxidant properties (Kim et al., 2017), as well as ascorbic acid (up to 0.25 g kg⁻¹). However, the antioxidant loading could be complemented by the immersion of MP fruit in solutions containing exogenous antioxidants (Aguayo et al., 2010). This process aims to interfere in oxidation reactions after MP and preserve endogenous phenolic compounds and ascorbic acid (Aguayo et al., 2010). When evaluating the concentrations of phenolic compounds in apples treated with antioxidant solutions (Figure 4a), it was observed that the EA 3 % maintained phenolic compounds concentrations constant during the nine days. In addition, it promoted the increase of these bioactive compounds throughout storage. The other treatments (LC, EA 1 % and EA 2 %), although resulting a lower amount of phenolics, remained within the values reported in the literature (50 mg.100 g⁻¹ FW to 380 mg.100 g⁻¹ FW) (Ceymann et al., 2012). The application of EA 3 % maintained the concentrations of

phenolic compounds. It occurs because EA, together with its stereoisomers, limits the production of *o*-quinones (Grant-Preece et al., 2013) by eliminating oxygen from the tissue before it reacts with the phenolic compounds (Clark et al., 2009, Bradshaw et al., 2011). In the absence of the antioxidant, PPO catalyses the phenolic compounds oxidation leading to the formation of undesirable pigments in the apple pulp (Zorzella et al., 2003). Oxidation, in addition to causing browning, may also result in loss of nutritional quality and provide flavor modifications. According to Son et al., (2001), the brown color intensity resulting from the PPO activity depends on the type of phenolic compounds involved.

The antioxidant activity was quantified based on the free radical scavenging activity (DPPH) (Figure 4b). The observed behavior for antioxidant activity was similar to that observed for total phenols (Figure 4a). EA 3 % resulted in the highest antioxidant activity, with a tendency to increase throughout storage, corresponding to a significant increase in free radical sequestration, from 444.68 mg Trolox.100 g⁻¹ FW (0 d) to 505.50 mg Trolox.100 g⁻¹ FW (9 d); an increase of 12 %. Aguayo et al., (2010) when working with the addition of antioxidants in 'Mariri Red' apples reported that to maintain a higher antioxidant activity than in the control treatment, a treatment with at least 6 % of calcium ascorbate was necessary. In the present study, EA 3 % maintained the antioxidant activity higher than in the TC as follows: 12.96 % (0 d), 14.11 % (3 d), 21.73 % (6 d) and 28.12 % (9 d). The increase in antioxidant activity over time can be attributed to the preservation of polyphenols, as shown previously. EA has antioxidant action due to its reducing

properties (Watanabe et al., 2014), acting in similar manner to AA (Clark et al., 2009). EA is considered safe for human consumption, low cost, well accepted by consumers and capable to increase the vitamin C content (Loan; Manzano, 1993),

even though it presents only 5 % of the AA vitamin activity. As for LC, its ability to preserve antioxidant activity in apples is attributed to its ability to sequester free radicals due to the presence of a thiol group (Altunkaya and Gökmen, 2008).

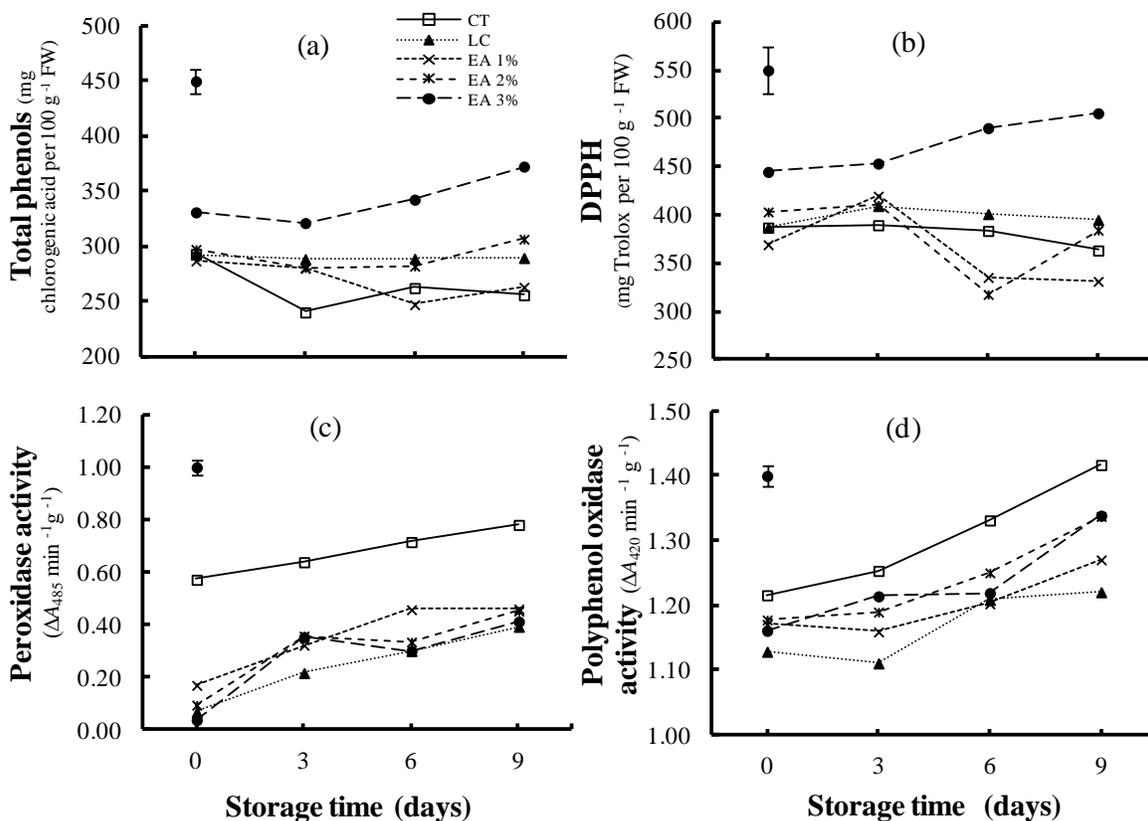


Figure 4. Total phenols (*LSD = 11.37), DPPH (*LSD = 24.20), peroxidase activity (*LSD = 0.026) and polyphenoloxidase activity (*LSD = 0.014) in minimally processed 'Royal Gala' apples stored for 0 d, 3 d, 6 d, and 9 d at 4.0 °C ± 1.0 °C, relative humidity of 90.0 % ± 5.0 %, after treatment with distilled water as a negative control treatment (CT); L-cysteine chloride (LC) 0.6 % (m.v⁻¹) as a positive control; and erythorbic acid (EA): EA 1 % (m.v⁻¹), EA 2 % (m.v⁻¹) and EA 3 % (m.v⁻¹). *Vertical bars indicate the Least Significant Difference (LSD) at p ≤ 0.05.

The control of PPO and POD activity after MP is important because these enzymes are involved in the darkening process; it occurs almost instantly after destruction of the cellular structure (Jang and Moon 2011). There was an increase in POD activity along the storage (Figure 4c), regardless of the application of antioxidants.

The remarkable difference was the lower activity of POD under LC and EA (1 %, 2 % and 3%) treatment in comparison to the activity under the CT. Considering that EA is an AA stereoisomer, the reduction in activity is in accordance with the results reported by Jang and Moon (2011), where the presence of AA effectively reduced POD

activity in MP apples. The reduced POD activity in AA treated fruits could be the result of lower oxidative stress on the fruit surface, due to the antioxidant nature of the molecular AA; it could also be a result of the formation of the POD-hydrogen donor complex (Saba and Sogvar, 2016).

The results observed for PPO activity (Figure 4d) are similar to those of POD activity. There was an increase in the PPO activity for all the treatments during the nine days of storage. Although there were oscillations in the enzymatic activity in apples treated with the antioxidant solutions, the PPO activity of these treatments was lower than the activity observed under the CT. Similar results were observed by Mirshekari et al., (2017) when working with MP 'Berangan' bananas treated with calcium propionate and chitosan. Even without major differences in POD and PPO activity among antioxidant treatments, LC and EA (1 %, 2 % and 3 %), the phenolic substrates were preserved as previously shown. This is an important result because one of the main reasons for the darkening of many fresh fruits and vegetables is the oxidation of phenolic substrates by PPO (Nguyen et al., 2003, Xing et al., 2010) and POD (Jang and Moon 2011). In addition to the antioxidant effect of LC and EA, the low pH of these solutions (2.22 and 1.62, respectively) may have contributed to the reduction of enzymatic activity. Tsouvaltzis and Brecht, (2017) when working with MP 'Russet Burbank' potatoes found that immersion of the MP tubers in H₂SO₄ pH 2.39 (< 0.04 %) reduced the PPO activity in comparison to the control treatment. According to these authors, this occurred because the optimum pH for the enzymatic activity (pH 5 to pH 7) was altered.

4. Conclusions

The use of erythorbic acid 3 % (m.v⁻¹) efficiently controls enzymatic browning and preserves physicochemical characteristics of minimally processed 'Royal Gala' apples under refrigeration for up to nine days.

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