



EFFECTS OF PROCESSING ON ESSENTIAL AND HEAVY METAL COMPOSITION OF POPULAR FISH SPECIES CONSUMED IN THE KARACHI COAST OF THE ARABIAN SEA

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ABSTRACT

This study analyzed three popular species of fish in raw and cooked (curried and fried) form, commonly consumed in Karachi coast and overall in Pakistan for their essential and heavy metal composition. The outcomes revealed that the content of toxic heavy metal (Hg) was observed in all raw samples particularly in raw Mulla fish (*Lethrinus nebulosus*) but heat treatment by frying and curry preparation of fish led to a decrease of Hg content in the muscles of all selected species especially in *L. nebulosus* after frying. While in *Acanthopagrus arabicus* Hg content was below detectable level. In this study the levels of Cd and Cr were observed at below detectable levels in all selected species except in *A. arabicus* where Cd tend to decrease after frying while the content of Cr slightly increase after both types of cooking. In the present study the essential metals namely Fe, Zn, Cu and Mn tend to increase in all fin fishes after cooking, especially in curried form. The above-mentioned result determined that these customary culinary practices of fish have an influence on their essential and heavy metal constituents. Furthermore, eating variety of fin fish species by applying different procedures of cooking is the finest attempt to attain better-quality of dietetic ways, minimizing mercury revelation and increasing chance to obtain vital elements.

1. Introduction

Toxic heavy metal pollution, which is obstinate and bio accumulative, progressively intimidates marine environment (Balkas et al., 1982; Bat et al., 2009; Bat, 2014). The metal pollutants in marine coastal system normally persist either in soluble or suspended form and finally tend to sink to the bottom or are taken up by the biota (Bat and Raffaelli 1998; Bat 2005; Khattak and Khattak, 2013; Bat and Özkan, 2015; Bat et al., 2015; Bat et al., 2017). Unfortunately, some heavy metals and their compounds are considered cancerous for humans and animals (Stanescu, 1998). Certain

metals for instance Cd, Pb, and Hg are extremely lethal even at very slight amounts (Bat, 2014; Bat, 2017). For example, Cd and Hg are familiar to cause kidney diseases, high blood pressure, cancer, hepatic dysfunction and harm reproductive ability while Pb can cause renal malfunction, liver mutilation, reduced hearing or produce mental obstruction, whereas at high intensities in women can result in a reduced conception period (Iwegbue, 2011). Hg, lethal effects have been emphasized when combined poisoning reported after consuming large amount of fish by people (Renzoni et al., 1998; Chen et al., 2002). Cu, Fe, Mn, and Zn are

crucial and compulsory for regular body activity like the synthesis of metalloproteinase. Although lack of these metals might lead to illness, unnecessary consumptions of these metals could initiate prolonged inflammatory sickness and a possible factor of tumor (Naughton and Petroczi 2008). Heavy metals can be hazardous to consumer's health (Diaconescu et al., 2012; Bat, 2014; Bat, 2017; Bat and Arici, 2018). Usually heavy metals store in fleshy tissue of fish and so, the intensities determined in their tissues are able to reveal the former exposure (Ashraf, 2005).

Karachi, Pakistan is a coastal metropolitan and thus facing industrialization difficulty. Hence, adjacent coastal regions of the Arabian Sea are getting an immense amount of unregulated industrial manure discharges that eventually disturb aquatic life (Jaffer et al., 1995; Tariq et al., 1998). Earlier studies have evidenced enhanced levels of these metals in fish belonging to south west coast of Pakistan (Tariq et al., 1998).

Usually fish species are consumed in cooked form while most of research studies information made from uncooked/raw products (Domingo, 2011). For the meantime heavy metal evaluation in raw products does not present the accurate calculations of these metals intake via seafood ingestion and consequently, it is essential to establish the accumulations of heavy metals in raw and cooked fish (Kalogeropoulos, 2012). Some study proven that the heavy metals concentrations of fish can also be changed by processing and hence, it is likely to decrease the toxic heavy metal concentration in fillets by selecting a proper technique of cooking (Kalay et al., 1999; Ersoy et al., 2006; Diaconescu et al., 2013). Many investigations have recorded a remarkable decline of the heavy metals in fish after cooking, while specific works mentioned rise in the metal amount. The works that revealed a reduction of toxic metals usage cooking processes subjected to cooking terms, such as time, temperature, and cooking medium (Morshey et al., 2015). Metal concentrations in fish are highly dependent on fish size (Ahmed et

al., 2016). According to Burger and Gochfeld (2011), the people must be well-informed prior to decide on which kind of fish is suitable to intake, how often and in what portion.

Though research studies are present on the heavy metal composition of raw fish, but studies to evaluate the essential and heavy metals composition of cooked fish in the Arabian Sea areas are rare. Furthermore, in Pakistan, almost none of the data exist on the impacts of traditionally processing on the essential and heavy metal concentrations in fish fillet. This study is consequently, an effort to evaluate the concentration of vital and toxic heavy metals in selected raw and traditionally proceed fish species usually consumed in Karachi.

2. Materials and methods

2.1. Materials

2.1.1. Collection of samples

Only the popular edible fish species of Karachi were included in this study. All fish were purchased from the local fish markets in Karachi city of Pakistan. Local, common, scientific names are presented in (Table 1). The selected culinary methods for selected fish species are common to most of the Pakistan and even in sub-continent.

Table 1. Common, Local and Scientific names of fish species included in the study

Common name	Local name	Scientific name
White Pomfret	Safeed Poplet	<i>Pampus argentus</i>
Arabian yellow-finned sea bream	Dhandya	<i>Acanthopagrus arabicus</i>
Spangled Emperor	Mulla	<i>Lethrinus nebulosus</i>

The fish were all captured from the coast of Karachi. On board they were put covered with ice and were later on shipped in a refrigerated truck to the central market. The selected fish species were purchased randomly during the period of August 2016 to February 2018. About 5 kg of each species of whole fish were captured, kept in icebox and immediately transferred to the PCSIR Laboratory of Karachi without delay,

where the identification and measurement of samples were taken.

For cooking purpose, vegetables like onion, tomato, garlic and mix spices, for coating of fish fillet, gram flour and sunflower cooking oil were purchased from local super market.

2.1.2. Sample preparation and cooking

Fish samples were washed with distilled water several times. The samples were cleaned as per usual cooking practices (scaling, beheading, gutting and removing the internal organs). Fillets were made for Yellow fin Bream, Mulla and Pomfret fishes. They were washed with water then fish fillet were divided into three groups, the first one is uncooked and the other two groups were cooked. Two different cooking methods were applied on fish species which are traditionally used on domestic and commercial level in local people. The frying of the marinated fish species was performed in a domestic non-stick pan (2-Litre capacity) at medium flame approximately for 15 minutes and fish curry was cooked with chopped vegetables and spices for 30 to 35 minutes in low to medium flame (Table 2).

Sunflower oil was used for frying and cooking in curry form. After cooking, samples cooled and kept below 4°C till analyzed in laboratory for essential and heavy metal concentration.

2.2. Methods

2.2.1. Heavy Metal Analysis

All chemicals used were of Analytical Regent (AR) Grade either of Merck or equivalent. Three fillet samples of each type raw or cooked fish selected at random and were dried in oven at 80°C for 3-4 hrs. The completely dried sample was homogenized with the help of pestle and mortar. For testing of Fe, Zn, Cu, Mn, Pb, Cr and Cd, 4-5 gm. of each sample in triplicate was weighed in a beaker and was soaked overnight in 5ml Conc. Nitric Acid. 10-15 ml deionized distilled water was added to the samples and the contents were heated at 60-80°C on a hotplate till samples were completely digested. Contents

were filtered through Whatman 41 in 25 ml volumetric flask and the volume was made with deionized distilled water. Sample blank was also prepared in the same manner, 5 ml Nitric acid was mixed with 10 – 15 ml deionized distilled water and heated on a hotplate for the same duration as for the samples. Samples were analyzed on Hitachi Z- 5000 Polarized Zeeman Flame/Graphite Furnace Atomic Absorption Spectrophotometer against standard curve. The instrument was handled as per manufacturer direction. Dilutions were made if needed to keep sample concentration between linear ranges of working curve.

For the analysis of Hg, 4-5 g of each sample in triplicate was reflux in 30 ml acid digestions mixture containing 1:1 nitric and sulfuric acid till contents are completely digested and all nitric acid is removed. Samples were diluted and solution was made up to 250 ml in volumetric flask with de-ionized distilled water. Sample blank was also made using similar method. Mercury was analyzed using same Hitachi Z 5000 Polarized Zeeman Flame/Graphite Furnace Atomic Absorption Spectrophotometer through cold vapor unit installed with the equipment against working curve. The instrument was handled as per manufacturer instructions. Dilutions were made if needed to keep sample concentration between linear ranges of working curve.

Metal contents were expressed as ppb for Hg, and ppm for Cu, Fe, Mn, Cd, Cr, Pb and Zn wet wt. of fresh fish.

2.2.2. Statistical analysis

The effect of different cooking methods on the proximate and heavy metal composition of selected fish was analyzed using standard deviation (SD).

The yearly quantity of fish consumed is 2 kg per person in 2006 (Food and Agriculture Organization of the United Nations 2009), which is same as 5.48 g/day for Pakistan. The EDI of metals was determined using the following equation (Bat and Arici, 2018).

$$EDI = C_{\text{metal}} \times W / b.w.$$

Where: C_{metal} is the concentration of metals in fish; W represents the per diem mean intake of fish; $b.w.$ is the body weight.

Table 2. Ingredients and traditional methods of preparation of fish species commonly consumed in Karachi, Pakistan

Species	Cooking process	Sample size (g)	Oil used (ml)	Ingredients	Method of preparation	Cooking time (min)
<i>Pampus argentus</i>	Fried	100	200	*	***	10-15
	Curry	100	60	**	****	25-30
<i>Lethrinus nebulosus</i>	Fried	100	200	*	***	10-15
	Curry	100	60	**	****	25-30
<i>Acanthopagrus arabicus</i>	Fried	100	200	*	***	10-15
	Curry	100	60	**	****	25-30

* Spices containing mainly, salt, red chili powder, turmeric powder, fresh garlic paste and gram flour for coating.

** Spices containing yellow mustard seed, onion, tomatoes, yoghurt, salt, red chili powder, turmeric powder and fresh garlic paste.

*** Wash the fillet/shrimp with salt and vinegar mix water then clean tap water, fillets coated with spices mix gram flour batter leave for 15 minutes in refrigerator for batter grip of coated ingredients then fry in moderate hot sunflower oil till brown.

**** Wash the fillet with salt and vinegar mix water then clean tap water, roast/brown the grind vegetables, yoghurt and spices in hot sun flower oil then add fillet in gravy and cooked in low to moderate heat till done.

3. Results and discussion

3.1. Heavy Metals Content

The recovery of the spiked metals was close to 97- 99% for all tested metals by proposed method. The mean concentrations (\pm SD) of heavy metals in raw, fried and curried form of fish muscles are given in Tables 3-5.

In *P. argentus* Fe, Zn and Cu contents were increased conspicuously in fried forms as compared to their raw forms. While as compare to raw samples the amounts of Mn and Hg were decreased in fried fish. Mn value was slightly higher in curried form whereas Fe value was low. The Hg content was decreased in fried form as compare to its curried and raw. The Cd, Cr and Pb contents were below the detectable level in raw and cooked forms of white Pomfret fish (Table 3).

In *L. nebulosus* the amount of Fe was remarkably increased in curried form than fried form as compare to its raw form. The amount of Hg is very higher observed in raw form while after frying it was surprisingly decreased

more than seven fold. The other heavy metal contents showed slightly increased after cooking. On the other hand Cd, Cr and Pb contents were below the detectable levels in raw and cooked forms (Table 4).

In *A. arabicus* Fe, Zn, Cu and Mn levels slightly increase in fried form while Cd amount decrease after frying and increase after cooked in curried form. The amount of Cr was slightly increased after both types of cooking as compare to raw form. The amounts of Hg and Pb were below the detectable levels in all raw and cooked samples of Bream fish (Table 5).

According to people well-being risk, the allowable weekly intakes were calculated by means of references for eatable tissues of fishes consumed by people. The EWI (Estimated Weekly Intake) and EDI (Estimated Daily Intake) levels showed in Tables 6-8, were estimated by assuming that a 70-kg person will consume 5.48 g fish per day which is even 38.36 g fish per week (see statistical analysis in Materials and Methods).

Table 3. Mean \pm SD of heavy metal concentrations given mg/kg except Hg ($\mu\text{g}/\text{kg}$) wet wt. in Raw and cooked *Pampus argentus* (white Pomfret fish)

Form	Fe	Zn	Cu	Mn	Hg	Cd	Cr	Pb
Raw	86 \pm 1.1	36 \pm 0.42	2.2 \pm 0.21	14 \pm 0.96	11 \pm 0.29	Nd*	Nd*	Nd*
Fried	108 \pm 2.12	95 \pm 1.04	18 \pm 0.21	9.5 \pm 0.78	5.9 \pm 0.15	Nd*	Nd*	Nd*
Curried	72 \pm 1.14	49 \pm 0.49	3.7 \pm 0.22	17 \pm 0.29	7.9 \pm 0.26	Nd*	Nd*	Nd*

Nd*= non detectable (below the detectable level)

Table 4. Mean \pm SD of heavy metal concentrations given mg/kg except Hg ($\mu\text{g}/\text{kg}$) wet wt. in Raw and cooked *Lethrinus nebulosus* (Mulla fish).

Form	Fe	Zn	Cu	Mn	Hg	Cd	Cr	Pb
Raw	34 \pm 0.5	16 \pm 0.21	1.4 \pm 0.17	0.7 \pm 0.15	144 \pm 1.21	Nd*	Nd*	Nd*
Fried	53 \pm 0.62	13 \pm 0.25	2.1 \pm 0.21	2.1 \pm 0.25	20 \pm 0.26	Nd*	Nd*	Nd*
Curried	74 \pm 1.36	19 \pm 0.26	1.7 \pm 0.23	3.9 \pm 0.29	47 \pm 0.98	Nd*	Nd*	Nd*

Nd*= non detectable (below the detectable level)

Table 5. Mean \pm SD of heavy metal concentrations given mg/kg except Hg ($\mu\text{g}/\text{kg}$) wet wt. in Raw and cooked *Acanthopagrus arabicus* (Arabian yellow fin bream fish).

Form	Fe	Zn	Cu	Mn	Hg	Cd	Cr	Pb
Raw	16.3 \pm 0.15	3.79 \pm 0.006	1.53 \pm 0.03	0.29 \pm 0.04	Nd*	0.008 \pm 0.002	0.138 \pm 0.002	Nd*
Fried	16.71 \pm 0.02	4.41 \pm 0.006	2.68 \pm 0.05	0.59 \pm 0.02	Nd*	0.006 \pm 0.003	0.188 \pm 0.011	Nd*
Curried	19.98 \pm 0.24	1.03 \pm 0.06	1.32 \pm 0.03	0.7 \pm 0.01	Nd*	0.023 \pm 0.003	0.156 \pm 0.001	Nd*

Nd*= non detectable (below the detectable level)

Table 6. Estimated Weekly Intakes (EWI) and Estimated Daily Intakes (EDI) of heavy metals in edible tissues of *Pampus argentus* (white Pomfret fish) from the local fish markets in Karachi city of Pakistan.

Metals	PTWI ^a	PTWI ^b	PTDI ^c	EWI ^d			EDI ^e		
				Raw	Fried	Curried	Raw	Fried	Curried
Fe	5.6	392	56	3.29896	4.14288	2.76192	0.47128	0.59184	0.39456
Zn	7	490	70	1.38096	3.6442	1.87964	0.19728	0.5206	0.26852
Cu	3.5	245	35	0.084392	0.69048	0.141932	0.012056	0.09864	0.020276
Mn	2-5	140-350	20-50	0.53704	0.36442	0.65212	0.07672	0.05206	0.09316
Hg	0.005	0.35	0.05	0.0004	0.00023	0.0003	0.00006	0.00003	0.00004
Cd	0.007	0.49	0.07	Not detectable (below the detectable level)					
Cr	0.0233	1.631	0.233	Not detectable (below the detectable level)					
Pb	0.025	1.75	0.25	Not detectable (below the detectable level)					

^aPTWI (Provisional Tolerable Weekly Intake) in mg/week/ kg body wt.^bPTWI for 70 kg adult person (mg/week/70 kg body wt.)^cPTDI (Permissible Tolerable Daily Intake) (mg/day/70 kg body wt.)^dEWI (Estimated Weekly Intake) (mg/week/ kg body wt.)^eEDI (Estimated Daily Intake) (mg/day/ kg body wt.)

Table 7. Estimated Weekly Intakes (EWI) and Estimated Daily Intakes (EDI) of heavy metals in edible tissues of *Lethrinus nebulosus* (Mulla fish) from the local fish markets in Karachi city of Pakistan.

Metals	PTWI ^a	PTWI ^b	PTDI ^c	EWI ^d			EDI ^e		
				Raw	Fried	Curried	Raw	Fried	Curried
Fe	5.6	392	56	1.30424	2.03308	2.83864	0.18632	0.29044	0.40552
Zn	7	490	70	0.61376	0.49868	0.72884	0.08768	0.07124	0.10412
Cu	3.5	245	35	0.053704	0.08056	0.065212	0.00767	0.01151	0.009316
Mn	2-5	140-350	20-50	0.026852	0.08056	0.149604	0.00384	0.01151	0.021372
Hg	0.005	0.35	0.05	0.0055	0.00076	0.00182	0.00078	0.00011	0.00026
Cd	0.007	0.49	0.07	Not detectable (below the detectable level)					
Cr	0.0233	1.631	0.233	Not detectable (below the detectable level)					
Pb	0.025	1.75	0.25	Not detectable (below the detectable level)					

^aPTWI (Provisional Tolerable Weekly Intake) in mg/week/ kg body wt.

^bPTWI for 70 kg adult person (mg/week/70 kg body wt.)

^cPTDI (Permissible Tolerable Daily Intake) (mg/day/70 kg body wt.)

^dEWI (Estimated Weekly Intake) (mg/week/ kg body wt.)

^eEDI (Estimated Daily Intake) (mg/day/ kg body wt.)

Table 8. Estimated Weekly Intakes (EWI) and Estimated Daily Intakes (EDI) of heavy metals in edible tissues of *Acanthopagrus arabicus* (Arabian yellow fin bream fish) from the local fish markets in Karachi city of Pakistan.

Metals	PTWI ^a	PTWI ^b	PTDI ^c	EWI ^d			EDI ^e		
				Raw	Fried	Curried	Raw	Fried	Curried
Fe	5.6	392	56	0.625268	0.6409	0.76643	0.08932	0.0916	0.10949
Zn	7	490	70	0.14538	0.1692	0.03951	0.02077	0.0242	0.0056
Cu	3.5	245	35	0.05881	0.1028	0.0506	0.0084	0.0147	0.0072
Mn	2-5	140-350	20-50	0.0111	0.0226	0.0269	0.0016	0.0032	0.0038
Hg	0.005	0.35	0.05	Not detectable (below the detectable level)					
Cd	0.007	0.49	0.07	0.0003	0.0002	0.0009	0.00004	0.00003	0.0001
Cr	0.0233	1.631	0.233	0.0053	0.0072	0.0059	0.0008	0.0010	0.0009
Pb	0.025	1.75	0.25	Not detectable (below the detectable level)					

^aPTWI (Provisional Tolerable Weekly Intake) in mg/week/ kg body wt.

^bPTWI for 70 kg adult person (mg/week/70 kg body wt.)

^cPTDI (Permissible Tolerable Daily Intake) (mg/day/70 kg body wt.)

^dEWI (Estimated Weekly Intake) (mg/week/ kg body wt.)

^eEDI (Estimated Daily Intake) (mg/day/ kg body wt.)

3.2. Discussions

According to recent reviews (Bat, 2014; Bat, 2017), results usually indicated that heavy metal absorptions remained minimal in the muscular tissues and maximal in the liver and gill. It has been revealed that destination tissues of toxic metals are metabolically active ones. Consequently, metal accumulation in destination tissues follow up greater amount compared to other tissues like the muscle, where metabolic activity is relatively weak (Kalay et

al., 1999; Roesijadi and Robinson, 1994; Serra et al., 1993; Langston, 1990; Heath, 1987).

Overall the toxic heavy metal such as Pb was not observed or below detectable level in raw and processed muscles of selected species of fish, it indicates that our traditional culinary practice did not effect on it. Musaiger and D'Souza (2008) observed that low level content of Pb i.e. (≥ 0.02) $\mu\text{g/g}$ in most of the cooked species of fish and shrimp of Arabian Gulf. In the present study the essential heavy metals such as Fe, Zn, Cu and Mn tend to increase in all fin

fishes after cooking, especially in curried form. It might be the reason of this increase is the usage of vegetables, yoghurt and citrus ingredients in traditional curry preparation (Musaiger 2006). The increases of Fe, Zn, Cu and Mn contents in fried form of finfish is may be the adding of gram flour batter coating on fillets. Gram flour is also rich in fibres, vitamin B-6, folate, thiamine, magnesium, phosphorus, potassium, manganese, iron, zinc and copper. It was also observed (Tawfik, 2013) the increase values of heavy metals after culinary practice, and was concluded that in the frying and marinated methods, the concentrations of metals increased. According to Bassey et al., (2014), the cooking methods produced remarkable raise in the concentrations of most metals compared to those of the uncooked samples. On the other hand, these culinary practices withal caused a decline in the amounts of metals in some fish species. This ways could be recognised to the interaction among the body size of the cooked fish, water loss, oil uptake and metal evaporation in the course of processing. However, according to Mitra et al. (2011), heat treatment by frying, boiling, steaming and curry preparation of fish lead to a reduction of the heavy metal amount in the muscle of all fish species. Devi and Sarojnalini (2012) observed the same as the decrease content of heavy metals in fry and curry form of Fish *Amblypharyngodon mola*.

In the present study the content of toxic heavy metal Hg was observed in all raw samples particularly in raw Mulla fish but heat treatment by frying and curry preparation of fish led to a decrease of Hg content in the muscles of all selected species especially in *L. nebulosus* after frying. However in *A. arabicus*, Hg content was below detectable level. According to Panichev and Panicheva (2016) cooking in sunflower oil might be the further example of thermal elimination of Hg from the fat fishes. They found 19.1% loss of Hg for rich in fat Yellowtail and only 5.9% for lean Cape hake fish.

In the present study the concentration of Cd and Cr was observed at below detectable levels in all selected species except in *A. arabicus*

where Cd tends to decrease after frying. The conflicting result has been observed by Bassey et al. (2014) where Cd tend to increase in *Polydactylus quadratifilis* after frying and grilling. Musaiger and D'Souza (2008) reported that the remained steady value i.e. $\geq 0.02 \mu\text{g/gm}$ of (Cd) for all the methods of cooking. Decrease in the metal levels during fish processing may be associated to the discharge of these contaminants by the loss of water as free salts, maybe in relation to soluble amino acids and uncoagulated proteins (Bryan and Hummerstone 1971). In *A. arabicus* the content of Cr slightly increase after both types of cooking. The same observed by Tawfik (2013) in fin fish species *Oreochromis niloticus*, *Mugil cephalus*, *Sardinops saga* and fried shrimp *Penaeus monodon* which showed increase values of Cr after frying and marinated form. Ahmed et al. (2015) pointed out that the Cr is very important for human diet because of its key role in insulin function and lipid metabolism. With respect to Western Australian Food and Drug legislations, the recommended maximal permissible amount of Cr is $5.5 \mu\text{g/g}$ (Plaskett and Potter, 1979). According to Bassey et al. (2014) the increase of Cr in the course of frying could be due to moisture loss and uptake of Cr from the oil during frying.

In this study the species wise differences was not constant for all the detected heavy metals. The obtained variation of metal amounts in several species counts on feeding habits (Romeo et al., 1999) environmental requirements, metabolism (Canli and Furness, 1993), their habitats (Canli and Atli, 2003; Tuzen and Soylak, 2006) and age, size and length of the species (Linde et al., 1998). The practice of culinary therefore plays a key role in altering the absorptions of heavy metals.

The several ways of cooking have a significant effect on the nutrient and heavy metal composition of fish. The changes are subject to on culinary circumstances (time, temperature and medium of cooking). It is found that the traditionally fried *P. argenteus* and curried form of *L. nebulosus* as a routine portion of the diet

would be useful because of its rich content of Fe, Zn and Cu. In the present study the increase in essential heavy metal contents could be due to the ingredients used in traditionally cooking practices. Consequently, it is probable to decrease the heavy metals in fish muscles by selecting an appropriate process of cooking. Atta et al. (1997) found that the Cd, Cu, Pb and Zn levels in *Tilapia nilotica* decreased on steaming and baking. However the reduction in these metal levels on baking was much higher than on steaming (Atta et al., 1997). In this study our results make public that heat energy has the significant role in separation of heavy metals from the fish. Hence in conclusion it can be encouraged that polluted fish by toxic heavy metal (such as Hg in *L. nebulosus*) may be consumed after traditionally culinary practice of the study. Thus, it is likely to reduce the metal levels in fish by choosing a proper method of cooking. Consequently, such fish flesh should only be consumed after cooking (Atta et al., 1997).

The tolerable weekly intake of the metals as PTWI (Provisional Tolerable Weekly Intake), are established by the Food and Agriculture Organization/World Health Organization (FAO/WHO) Joint Expert Committee on Food Additives (JECFA). PTWI is the maximal level of a pollutant to which a person can be exposed per week over a lifespan with no health risk effects (National Academy of Science, 1989; WHO, 1996; Council of Europe, 2001; FAO/WHO, 2010; EFSA, 2010; EFSA, 2012). EDI levels of metals for a person (mg/70 kg body weight) consuming 38.36 g seafood/week were estimated using the mean \pm SD metal levels (see Tables 6-8) for *P. argentus*, *L. nebulosus* and *A. arabicus*. Intake estimates were expressed as per unit body weight (mg/kg body wt. /weekly and daily). EDI values were calculated from EDI values.

4. Conclusions

In this study, the calculated EWIs and EDIs of the metals are under the permitted PTWIs and

PTDIs and showed no hazard consequences to the consumers.

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