journalhomepage:http://chimie-biologie.ubm.ro/carpathian_journal/index.html

VOLATILE COMPONENTS OF STRAWBERRY JAM

Irvna Zamorska^{*1}, Volodvmvr Zamorskvi²

¹Department of Technology Storage and Processing of Fruits and Vegetables, Uman National University of Horticulture,

²Department of Fruit Growing and Viticulture, Uman National University of Horticulture, zil197608@gmail.com

https://doi.org/10.34302/crpjfst/2019.11.2.7

Article history:	ABSTRACT			
23 April 2018	studied to define the content of aromatic volatiles using the methods of			
Accepted:	highly efficient liquid chromatography. Volatiles contain a considerable			
10 March 2019	amount of acids (65.6-76.8%), a small amount of furanone (8.5-14.6%) and that of aldehydes (3.4-10.8%). The share of esters in jams exceeds 0.7-3.1%			
Keywords: Compounds; Flavor; Jam; Strawberry; Volatiles	of the total volatile amount. Typical compounds for strawberry jam flavor are hexanoic (caproic) acid, hexadecanoic acid, 2-ethyl hexanoic (capronic) acid, trans-cinnamic acid, linoleic acid, furil hydroxy methylketone, 2,5- dimethyl-4-methoxy-3(2H)-furanone (mesifurane), furfural, 5- hydroxymethylfurfural, vanillin. As to aroma activity furanone derivatives dominate: 2.4-dioxy-2.5-dimethyl-3(2H)-furan-3-one, 2.5-dimethyl-4- methoxy-3(2H)-furanone (mesifurane) Ta 2.5-dimethyl-4-hydroxy-3(2H)- furanone (furaneol); they add sweet caramel scents to jams. For strawberry jams of cultivar "Polka", ethyl 2-methylbutanoate, decanal are active components which add fruity and grassy scents, strawberry jam made of			
	(fruity, sweet) and linalool (sweet and floral scent).			

1.Introduction

Garden strawberry is the most popular and valuable berry crop due to its high flavoring characteristics, fast and early ripening, notdemanding to growing conditions, high yielding capacity and high economic efficiency of growing. Strawberries are rich in sugars, organic acids, vitamins, phenol compounds, mineral substances, they have well-expressed typical flavor; they are in constant demand among consumers due to their high gustatory properties (Markovskyi et al., 2008). Strawberries are consumed both as fresh and as juices, drinks, wines, puree, stewed fruits, jams (Amaro et al., 2012).

Strawberry flavor is a combination of esters, aldehydes, ketones, ethyl, lactones, terpenic

(Pérez et al, 1996; Jetti, 2005; Kim et al., 2013). Strawberry taste develops during ripening and it changes during storage (Zabetakis et al., 1997; Forney, 2000) and processing for canned products (Lambert, et al. 1999). One of the most popular processing products made of strawberry is jam which is due to high organoleptic properties, availability consumers and long shelf-life of the product. To

compounds, furanones (Larsen et al., 1992;

Larsen et al., 1992, Forney et al., 2000, Kafkas et al., 2005). Esters are the main components

(25-90% of the total amount), aldehydes and

furanones - 50% (Larsen et al., 1992; Larsen et

al., 1992). The latter add fruity and floral green

and sweet or caramel flavors to strawberries

get strawberry jam with well-felt strawberry

for

flavor it is advisable to use the most fragrant berries (Lesschaeve *et al.*, 1991; Suutarinen *et al.*, 2002).

The main volatile compounds of strawberry jams belong to the classes of acids, alcohols and esters (Barron D. *et al.*, 1990; Kimura *et al.*, 1994.); they have natural origin and can be formed as a result of heat treatment (Sloan *et al.*, 1969). Thus, due to high temperatures of the treatment, changing sugar into caramel and Maiyar reaction, the product gets boiled, burned and caramel taste (Avasoo *et al.*, 2011), whereas green and fruity flavor, typical for fresh berries, becomes less expressed (Ozcan *et al.*, 2011). High furanol concentrations add typical caramel and sweet flavors to strawberry jams (Lesschaeve *et al.*, 1991; Pérez *et al.*, 1996).

However, not enough information is available in scientific literature concerning the content and composition of volatiles of jam made of strawberries of some pomological cultivars.

The purpose of our research was to identify the content and composition of volatiles of jam made of strawberry cultivars Ducat, Honey and Polka.

2. Materials and methods

2.1. Materials

The work was done in 2013-2014 with the berries of varieties "Ducat", "Honey", "Polka" in the laboratory of the department of the technology of storage and processing of fruits and vegetables at Uman national university of horticulture and at the experimental center of foodstuff quality control at the National institute of grape and wine "Magarach" (Ukraine).

2.2. Technological process

Strawberries were harvested at a technical stage of ripening, sorted by quality, cleaned and washed. Jams were made of the prepared berries according to current technological instruction (1992) adding pectin in concentration of 0.3%. Jam was boiled until the content of dry soluble substances reached 62%, packed in glass jars (250 cm³). Jam was made of each pomological

cultivar in five replications. The product was kept for six months at 20 $^{\circ}$ C.

2.3. Main instruments and equipment

Agilent Technologies 6890 chromatograph with mass-spectrometric detector 5973 and chromatographic capillary column DB-5 - internal diameter 0.25 mm and length 30 M – was used to identify volatiles of finished jam.

2.4. HPLC analysis on volatile components

A sample (0.75 g) was put in a 2-ml vial, and internal standard was added. Trydekan (50 mkg per sample) was used as an internal standard. To extract volatile substances, 0.5 ml of chloride methylene was added, the exposure time was 24 hours. The vial was carefully shaken several times. The extract received was taken with a micro-syringe, put into a 2-ml vial and digested in the flow of specifically clean nitrogen to the volume of 50 mkl. The concentrate was chromatographed.

A sample was injected into a chromatographic column in splitless mode, i.e., the flow was not divided; this allowed to eliminate losses and to increase the sensitivity of chromatography method considerably (10-20 times). The speed of the sample injection was 1.2 ml/min, and it took 0.2 min.

To identify the components, a library of mass spectrums NIST05 and WILEY 2007 with the total number of spectrums more than 470000 in combination with identification programs AMDIS i NIST was used.

The method of internal standard was used for quantitative calculations.

The calculation of the component content was made using the equation where:

$$C = K_1 \times K_2$$

C-volatile component content, mg/kg,

$$K_1 = \frac{\Pi_1}{\Pi_2}$$

 $\Pi 1$ – peak area of the substance studied, $\Pi 2$ – peak area of the standard;

$$K_2 = \frac{50}{M}$$

50 - mass of the internal standard (mkg), introduced into a sample, M – a sample (g).

2.5. Statistical Analysis

Statistic analysis was made using StatSoft STATISTICA 6.1.478 Russian, Enterprise Single User (2007).

3.Results and discussions

38 components were identified in volatile concentration in strawberry jams made of the varieties studied: esters, aldehydes, ketones, furanones, acids, aroma compounds, lactones, terpenic compounds. The concentration of volatiles in strawberry jams was 12.1-33.54 mg/kg depending on the variety (Table 1). The most meaningful shares are: acids -65.6-76.8%, furanes - 8.3-14.6% and aldehydes - 3.4-10.8% (Fig.1). The share of esters in jams exceeds 0.7-3.1% of the total volatile content. It is important to mention that in strawberry jams made of Polka cultivar the share of esters and aldehydes is much higher: 3.1 and 10.8%, that of furanes and acids, on the contrary, is the lowest -8.3 and 65.6% which proves strong expression of scents typical for fresh strawberries.

Characteristic compounds for strawberry jam flavor made of the studied cultivars are hexanoic (caproic) acid (0.84-6.89 mg/kg), which is 6.9-22.9 % of the total volatile amount depending on their quantity for each cultivar, hexadecanoic acid (2.5-12.4%), 2- ethyl hexanoic (capronic) acid (3.1-10.7%), transcinnamic acid (17.5-25.3%), linoleic acid (0.3-7.2%), furil hydroxy methylketone (3.1-6.0%), 2.5-dimethyl-4-methoxy-3(2H)-furanone

(mesifurane) (7.4-13.7%), furfural (0.8-3.1%), 5-hydroxymethylfurfural (0.8-5.2%), vanillin (0.2-0.8%).

The availability of furfural (0.8-3.1%), 5hydroxymethylfurfural (0.8-5.2%) 5methylfurfural (0.7%) in strawberry jams indicates non-fermentative darkening during thermal treatment (Barren *et al.*, 1990; Kimura *et al.*, 1994).

High content of 2-heptenal -0.9% of the total volatile content and hexanol (1.9%) were

found in strawberry jam made of Polka cultivar, and as to cultivars "Ducat" and "Honey" – high content of 2-methylbutyric acid (6.3-6.6%) γ , and 2.5-dimethyl-3(2H)-furanone (0.4-1.1%). According to Schwab (2013) 2.5-dimethyl-3 (2H)-furanones is synthesized via sets of fermentative changes in fruits.

 γ -decalactone – 0.55 and 1.73 mg/kg which is 1.8-5.2% – was found in strawberry jams (cultivars "Ducat" and "Honey"). This volatile compound adds "fruity", "sweet" and "peachy" scents (Ulrich *et al.*, 2007).

Small amounts of 2H -pyran-2.6(3H)-dion and 3.5-hydroxy-2-dimethyl-4H- pyran-4 – (0.09-0.66 mg/kg) which, depending on the cultivar, is 0.5-0.9% of the total volatile content in jams, were found; and according to (Barren *et al.*, 1990) they are the products of Maiyar reaction resulted from the reaction of glucose with glutamic acid, glycine, butylamine, lysine, hydroxypyroline and/or phenylalanine (amino acids).

Terpenic compounds of strawberry jams are presented by small amounts of limonene (0.1 mg/kg, which is 0.8%) and α -terpineol (0.2-1.3%), they were found in fresh berries by Bianchi *et al.* (2014); these compounds add aromatic scent to fresh berries (Ulrich *et al.*, 2007, Bianchi *et al.*, 2014); oxyde bisabolol A (0.1-0.3%), trans - linalool oxyde (0.1-0.3%), cis - linalool oxyde (0.3%), however no data concerning their presence in fresh strawberries is available. Despite a large volatile amount, strawberry jam flavor is developed under the effect of their small amount.

To determine the share of each compound in the flavor, its activity is defined by dividing substance concentration on its threshold concentration (OAV = concentration (ppbv)/threshold value (ppbv) (Table 2) (Rothe *et al.*, 1963, Kim *et al.*, 2013). If the result of OAV exceeds 1, it proves the contribution of a component to the flavor.

V-1-41	Variety				
volatiles	Polka	Ducat	Honey		
Esters		·	· •		
Methyl butanoate	0.14	0.03	-		
Ethyl butanoate	0.06	0.15	0.09		
Ethyl crotonate	0.02	-	-		
Ethyl 2-methyl butanoate	0.02	-	-		
Ethyl capronate	0.01	-	-		
3,4-dihydropyran	0.13	0.10	0.12		
Total esters	0.38	0.28	0.21		
Aldehydes		·	·		
Benzaldehyde	0.04	0.10	0.14		
Trans-2-heptenal	-	-	0.14		
Hexanal	0.06	0.04	0.03		
2-heptenal	0.11	-	-		
Decanal	0.01	-	-		
Furfural	0.38	0.38	0.23		
5- hydroxymethylfurfural	0.63	0.66	0.24		
Undecenal	0.05	-	-		
2-decenal	-	0.07	-		
Vanillin	0.03	0.20	0.23		
5- methylfurfural	-	0.24	-		
Total aldehydes	1.31	1.69	1.01		
Aromatic compounds			4		
Hexanol	0.23	-	-		
2H -pyran-2,6(3H)-dion	0.09	0.50	0.27		
3,5-hydroxy-2-dimethyl-4H- pyran-4-on	-	0.66	-		
Total aromatic compounds	0.32	1.16	0.27		
Acids		1	1		
2-Methylbutyric acid	_	2.22	1.89		
Octanoic acid	0.10	0.66	0.40		
Nonanoic acid	-	0.09	-		
Hexanoic (caproic) acid	0.84	4.85	6.89		
Tetradecanoic acid	0.32	0.24	0.36		
Palmitoleic acid	0.38	0.21	0.48		
Hexadecanoic acid	1.50	0.85	1.37		
2- ethyl hexanoic (capronic) acid	1.29	2.11	0.95		
Trans-Cinnamic Acid	2.11	7.74	7.64		
Dodecanoic acid	0.07	0.07	0.12		
Pentadecanoic acid	0.21	0.10	0.26		
Linoleic acid	0.87	0.85	0.10		
Octadecanoic acid	0.24	0.11	0.21		
Cis-Cinnamic acid	-	1.73	2.10		
Oleic acid	_	0.33	0.38		
Total acids	7.93	22.16	23.15		

87

Zamorska	and Zamorskyi	/Carpathian	Journal of	^f Food Science	and Tec	hnology 2	2019,11(2),	84-92
----------	---------------	-------------	------------	---------------------------	---------	-----------	-------------	-------

Lactones			
γ- Decalactone	-	1.73	0.55
Butyrolactone	0.12	0.38	-
γ-Caprolactone	0.08	-	-
δ- Caprolactone	-	0,04	-
Total lactones	0.20	2.15	0.55
Ketones			
2-acetylfuran	0.04	-	-
Furil hydroxy methylketone	0.73	1.13	0.95
Total ketones	0.77	1.13	0.95
Furan derivatives			
2,4-dioxy-2,5-dimethyl-3(2H)-furan-3-one	0.12	0.10	0.10
2,5-dimethyl-4-methoxy-3(2H)-furanone	0.80	4.61	236
(mesifurane)	0.89	4.01	2.30
2,5-dimethyl-4-oxy-3(2 <i>H</i>)-furanone	-	0.03	0.56
2,5-dimethyl-3(2 <i>H</i>)-furanone	-	0.15	0.34
Total furan derivatives	1.01	4.89	3.36
Terpenes		-	
Linalool	-	-	0.08
α-Terpineol	0.03	-	0.40
Limonene	0,10	-	-
Oxyde bisabolol A	0.04	0.05	-
Trans - linalool oxyde	-	0.03	0.08
Cis - linalool oxyde	-	-	0.08
Total terpenes	0.17	0.08	0.64
Total amount	12.09	33.54	30.14
LSD ₀₅		0.06	





Figure 1. Volatile components of strawberry jams made of various cultivars, % of the total content

The calculation of volatile OAV of strawberry jams showed that furanone derivatives dominated: 2.4-dioxy-2.5-dimethyl-3(2H)-furan-3-one, 2.5-dimethyl-4-methoxy-3(2H)-furanone (mesifurane) and 2.5-dimethyl-4-hydroxy-3(2H)-furanone (furaneol); they add sweet, caramel scents to strawberry jams. Vanilla and hexanal are also active compounds, they are typical for vanilla and fresh grassy scents (Barren et al., 1990); 2-methylbutanoic acid, which adds sour-sweet taste, is active in strawberry jams made of "Ducat" and "Honey" cultivars.

For strawberry jams, ethyl 2-methylbutanoate, decanal along with furanes are active components which add fruity and grassy scents to strawberry jams made of cultivar "Polka". 2decenal and γ - decalactone, 'Honey' – γ decalactone (fruity, sweet) and linalool (sweet and floral scent) make an important contribution to the flavor of strawberry jam made of cultivar "Ducat".

Volatiles	Theshold, mg/kg	Activity of flavor volatile components (OAV)			
		Polka	Ducat	Honey	
Methyl butanoate	0.06	2.3	0.5	-	
Ethyl butanoate	0.018	0.3	8.3	5.0	
Ethyl crotonate	NA ¹	-	-	-	
Ethyl 2-methylbutanoate	0.0001	200	-	-	
Ethyl capronate	NA	-	-	-	
Benzaldehyde	0.35	0.1	0.3	0.4	
3,4-dihydropyran	NA	-	-	-	
Trans-2-Heptenal	0.013	-	-	10.8	
Hexanal	0.0045	13.3	8.9	6.7	
2-Heptenal	NA	_	_	_	

Table 2. Activity of volatile components of strawberry jam flavor (OAV)

Decanal	0.0001	100	-	-
Furfural	3.0	0.13	0.13	0.08
Undecenal	0.005	10	_	-
2-decenal	0.0003	-	233	-
Vanillin	0.02	1.5	10	11.5
5- hydroxymethylfurfural	NA	-	-	-
5- methylfurfural	NA	-	-	-
Hexanol	2.5	0.09	-	-
2H -pyran-2,6(3H)-dion	NA	-	-	-
3,5-hydroxy-2-dimethyl-4H- pyran-4-on	NA	-	-	-
2-Methylbutanoic acid	0.25	-	8.9	7.6
Octanoic acid	0.910	0.1	0.7	0.4
Nonanoic acid	3	-	0,03	-
Hexanoic (caproic) acid	1.0	0.8	4.9	6.9
2- ethyl hexanoic (capronic) acid	NA	-	_	-
Trans-Cinnamic Acid	NA	-	_	-
Dodecanoic acid	10	-	-	-
Pentadecanoic acid	NA	-	-	-
Linoleic acid	NA	-	_	-
Octadecanoic acid	20	0.01	0.006	0.01
Cis-Cinnamic acid	NA	-	-	-
Oleic acid	NA	-	-	-
Tetradecanoic acid	10	0.03	0.02	0.04
Palmitoleic acid	NA	-	-	-
Hexadecanoic acid	NA	-	-	-
γ- Decalactone	0,01	-	173	55
Butyrolactone	NA	-	-	-
γ-Caprolactone	NA	-	-	-
δ- Caprolactone	NA	-	-	-
2-acetylfuran	10	0.004	-	-
Furil hydroxy methylketone	NA	-	-	-
2,4-dioxy-2,5-dimethyl-3(2 <i>H</i>)-furan-3-one	0.00004^2	3000	2500	2500
2,5-dimethyl-4-methoxy-3(2 <i>H</i>)-furanone	0.000032	20667	153667	78667
(mesifurane)	0.00003	29007	155007	/800/
2,5-dimethyl-4-hydroxy-3(2H)-	0.00004^{2}		750	14000
furanone (furaneol)	0.00004	-	750	14000
2,5-dimethyl-3(2 <i>H</i>)-furanone	NA	-	-	-
Linalool	0.006	-	-	13.3
α-Terpineol	0.330	0.09	-	1.2
Limonene	0.01	10	-	-
Oxyde bisabolol A	NA	-	-	-
Trans - linalool oxyde	NA	-	-	-
Cis - linalool oxyde	NA	-	-	-

¹NA – not available. Threshold levels of compounds (in water) were obtained from the flavor base of Leffingwell & Associates.

²Siegmund B., Bagdonaite K., Leitner E. (2010)

Fruity and floral, green and sweet or caramel scents are emphasized in the aroma of fresh strawberries. Esters, ethyl acetate, butyl acetate, methyl butanoate, ethyl butanaote, ethylisovalerate, methyl hexanoate and ethyl hexanaote add fruity and floral scents to strawberry flavor. Hexanal, trans-2-hexenal, 2hexenal, hexanol, cis-3-hexen-1-ol, hexyl acetate add green scents, and furaneols - sweet, caramel ones (Pérez et al., 1996; Jetti, 2005; Kim et al., 2013). Having analyzed volatile activity data of strawberry jams, it has been established that typical scents for them are sweet and caramel ones due to high furaneol activity, also there are vanilla, fruity and fresh grassy scents. The flavor of strawberry jams made of "Polka" cultivar is characterized with fruity and grassy scents, that of "Ducat" cultivar - fruity and sweet scents, and for "Honey" cultivar sweet and floral scents.

4. Conclusions

The flavor of strawberry jams made of "Polka", "Ducat" and "Honey" cultivars consists of a complex mixture of compounds, the most active among them are furanone ones: 2.4-dioxy-2.5-dimethyl-3(2H)-furan-3-one, 2.5-dimethyl-4-methoxy-3(2H)-furanone

(mesifurane) and 2.5-dimethyl-4-hydroxy-3(2H) furanone(furaneol). Ethyl 2methylbutanoate, hexanal, decanal, 2decenal, vanillin and γ - decalactone make a great contribution to the flavor.

Sweet, caramel scents with vanilla and fresh grassy scents are very typical for strawberry jams: the flavor of strawberry jams made of "Polka" cultivar is characterized with fruity and grassy scents, that of "Ducat" cultivar – fruity and sweet scents, and for "Honey" cultivar – sweet and floral scents.

5. References

Amaro, L. F., Soares, M. T., Pinho, C., AlmeidaI. F., Ferreira I. M. P. L. V. O. Pinho O. (2012). Influence of cultivar and storage conditions in anthocyanin content and radical-scavenging activity of strawberry

jams. World Academy of Science, Engineering and Technology, Vol. 69.

- Avasoo, M. & Johansson, L. (2011). Evaluation of thermal processing technologies for strawberry jam.
- Barron, D. & Etiévant, P. X. (1990). The volatile constituents of strawberry jam. *Zeitschrift für Lebensmittel-Untersuchung und Forschung*, 191(4-5), 279-285.
- Bianchi, G., Lovazzano, A., Lanubile A. & Marocco A. (2014). Aroma Quality of Fruits of Wild and Cultivated Strawberry (Fragaria spp.) in Relation to the Flavour-Related Gene Expression. *Journal of Horticultural Research*, 22(1), 77-84.
- Forney, C.F., Kalt, W. & Jordan, M.A. (2000). The composition of strawberry aroma is influenced by cultivar, maturity, and storage. *HortScience*. 35(6), 1022-1026.
- Jetti, R.R., (2006). Fruit Quality Evaluation of Strawberries (Fragaria ananassa) Grown in California and Oregon. Oregon State University.
- Kafkas, E., Kafkas, S., Koch-Dean, M., Schweb W., Larkov O., Lavid N., Bar E., Ravid U. & Lewinsohn E. (2005). Comparison of methodologies for the identification of aroma compounds in strawberry. *Turkish Journal of Agriculture and Forestry*. 29(5), 383-390.
- Kim, Y.H., Kim, K.H., Szulejko J.E. & Parker D. (2013). Quantitative analysis of fragrance and odorants released from fresh and decaying strawberries. *Sensors*. 13(6), 7939-7978.
- Kimura K., Ida M., Yosida Y., Ohki K., Fukumoto T. & Sakui N. (1994). Comparison of keeping quality between pressure-processed jam and heat-processed jam: Changes in flavor components, hue, and nutrients during storage. *Bioscience, biotechnology, and biochemistry*, 58(8), 1386-1391.
- Lambert, Y., Demazeau, G., Largeteau, A. & Bouvier J. M. (1999). Changes in aromatic volatile composition of strawberry after high pressure treatment. *Food Chemistry*, 67(1), 7-16.

- Larsen, M. & Poll, L. (1992). Odour thresholds of some important aroma compounds in strawberries. Zeitschrift für Lebensmittel-Untersuchung und Forschung. 195(2), 120-123.
- Larsen, M., Poll, L. & Olsen, C.E. (1992). Evaluation of the aroma composition of some strawberry (Fragaria ananassa Duch) cultivars by use of odour threshold values Zeitschrift für Lebensmittel-Untersuchung und Forschung. 195(6), 536-539.
- Lesschaeve, I., Langlois, D. & Etiévant, P. (1991). Volatile compounds in strawberry jam: influence of cooking on volatiles. *Journal of food science*, 56(5), 1393-1398.
- Markovskyi, V.S., Bakhmat, M.I. (2008) Small berries in Ukraine: a study guide. Kamianets-Podilskyi: PB "Medobory-2006". 200 p. [in Ukrainian].
- Ozcan, G. & Barringer, S. (2011). Effect of enzymes on strawberry volatiles during storage, at different ripeness level, in different cultivars, and during eating. *Journal of food science*, 76(2), 324-333.
- Pérez, A.G., Olías, R., Sanz, C. & Olías, J.M. (1996). Furanones in strawberries: evolution during ripening and postharvest shelf life. Journal of *Agricultural and Food Chemistry*. 44(11), 3620-3624.
- Rothe, M. & Thomas, B. (1963). Aromastoffe des brotes: Versuch einer auswertung chemischer geschmacks analysen mit hilfe des schwellenwertes. Z. Lebensm. Unters. Forsch. 119, 302- 310. (in German).
- Schwab, W. (2013). Natural 4-Hydroxy-2, 5dimethyl-3 (2H)-furanone (Furaneol®). *Molecules*, 18(6), 6936-6951.
- Siegmund, B., Bagdonaite K. & Leitner E. (2010). Furaneol and mesifuran in strawberries - an analytical challenge. *Expression of Multidisciplinary Flavour Science*, 537-540.
- Sloan, J. L., Bills, D. D. & Libbey, L. M. (1969). Heat-induced compounds in strawberries. Journal of Agricultural and *Food Chemistry*, 17(6), 1370-1372.
- Suutarinen, M. (2002). Effects of prefreezing treatments on the structure of strawberries

and jams. VTT Technical Research Centre of Finland.

- Ulrich, D., Komes, D., Olbricht, K. & Hoberg, E. (2007). Diversity of aroma patterns in wild and cultivated Fragaria accessions. *Genetic Resources and Crop Evolution*, 54(6), 1185-1196.
- Zabetakis, I. & Holden, M. A. (1997). Strawberry flavour: analysis and biosynthesis. *Journal of the Science of Food and Agriculture*, 74(4), 421-434.
- Book of technological instructions how to make canned goods (1992). "Konservplodovoshch". Moscow: Publishing house "Petit", volume II. Canned fruits. Part. I. [in Ukrainian].