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## MANAGING QUALITY OF AROMATIZED WINE PREPARED BY CO-FERMENTATION OF GRAPE MUST AND BY-PRODUCTS OF ESSENTIAL ROSE OIL INDUSTRY

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Article history:	ABSTRACT
Received:	Aromatized rosé wines with addition of essential rose oil industry wastes
2 February 2021	during fermentation of grape must were prepared. Six variants: W1-W6 with
Accepted:	added 0.05%, 0.1%, 0.25%, 0.5%, 1%, and 2% Rosa damascena Mill. waste,
25 July 2021	respectively, and control wine were prepared. Slight differences in the color
Kevwords:	shades were observed: the lower the added rose waste, the more intensive
Aromatized wine:	peony color was obtained and this observation was confirmed with the
Rosé:	increase of the hue angle value $-46.21\pm0.84$ for the control and $54.95\pm0.70$
Mavrud;	for the W6. The polyphenol content increased significantly from
Rosa damascena Mill.;	$355.01\pm10.14$ to $576.08\pm12.08$ µmol GAE L <sup>-1</sup> for the control and W6,
Co-fermentation.	respectively. The major phenolic acids determined were 3,4-dihydroxy
,	benzoic (up to $65.1\pm1.1 \text{ mg L}^{-1}$ in W6), gallic (up to $25.9\pm0.9 \text{ mg L}^{-1}$ in W6)
	and chlorogenic acid (up to $11.7\pm0.6 \text{ mg L}^{-1}$ in W5). The GC-FID analysis
	revealed slight increase of higher alcohols for W5 and W6. β-Caryophyllene,
	β-citronellol, phenethyl alcohol, rose oxide, and geraniol content increased
	significantly compared to control. The sensory evaluation revealed most of
	the panelists preferred W1 and W2 although some of the testers liked better
	the variants with higher amounts of added waste. The results suggested that
	rose waste successfully could be utilized for preparation of new aromatized
	wines with distinctive rose aroma

#### **1.Introduction**

The wine is among the most popular and produced worldwide low alcoholic beverages. The fundamental factors determining the wine quality are geographical region, climate conditions, soils, grape variety, stage of ripeness, yeasts, as well as, vinification (Cioch-Skoneczny *et al.*, 2021; Nardi *et al.*, 2018). The wine aroma is among the most important factors for the wine quality and acceptance (Nardi *et al.*, 2018). The major contributors for the formation of aroma bouquet are the yeast fermentation of grape must and skin contact time (Cabaroglu and Canbas, 2002). Furthermore, the aroma could be modulated by addition of other flavoring substances and these beverages are categorized as aromatized wines. The aromatized wines, according to Regulation 251/2014 of the European Parliament and the Council, are

defined as wines with organoleptic characteristics achieved by addition of natural flavoring substances and/or herbs and spaces, including their extracts, and/or flavor products, and combination thereof. Different flavoring materials were used: wormwood, dwarf gentians (Gentianella sp.), mint, cinnamon, green cardamom, elderberry, nutmeg, rosemary, juniper, Hypericum sp., clove, flat-leaved vanilla, etc. The utilization of agricultural byproducts is a rare practice but some aromatized wines exist, i.e. St. Raphael's aperitif wine prepared with bitter orange peels (Buglass, 2011). By-products from the olive oil industry were used in an attempt for replacing sulfur dioxide in wine models (Ruiz-Moreno et al., 2015), and overripe seeds from white grape byproducts were added during red wine fermentation in order to investigate the effect on wine color and phenolic substances (Rivero et al., 2017). Attempts for preparation of aromatized wines with addition of essential oil industry main products were made but problems with solubility and separation of the oils and wine during storage were observed. The literature survey suggested, to the best of our knowledge, that no attempts for preparation of aromatized wines with addition of by-products of the essential-oil industry, which emblematic and widespread in some European and Asian countries (Bulgaria, France, Turkey, Iran, China, etc.), were described. By-products of the rose oil-industry are usually not further utilized and are discarded, although the waste could serve as a valuable raw material for obtaining of biologically active substances (Slavov et al., 2017). For this reason, based on the above-mentioned observations, literature survey and experimental data, the present study aimed to investigate the possibility for preparation and managing quality of aromatized wines with addition of rose oil industry waste in the course of Mavrud must fermentation.

## 2. Materials and methods 2.1.Materials

### 2.1.1.Samples and reagents

The *Rosa Damascena* Mill. waste was provided by EKOMAAT Ltd. distillery (Mirkovo, region of Sofia, Bulgaria; 2016 harvest; waste obtained from certified bio roses was used). The grape used for wine preparation was *Vitis vinifera* L. cv Mavrud (Brestovica, region of Plovdiv, Bulgaria; 2016 harvest) with 23.8% sugars and 8.2% titratable acids. The Lallzyme cuvée blanc and the yeast strain Lalvin D47 were obtained from Lallemand (France). The Polymust press and bentonite were obtained from Laffort (France).

Acetonitrile, acetic acid, dichloromethane, sodium acetate, pyridine, N,O-Bis-(trimethylsilyl)-trifluoroacetamide, gallic acid, 3,4-dihydroxy benzoic acid, chlorogenic acid, caffeic acid, p-coumaric acid, ferulic acid, sinapic acid, rosmarinic acid, cichroric acid and cinnamic acid were obtained from Sigma-Aldrich (USA). The DPPH (2,2-diphenyl-1picrylhydrazyl) was from Merck (Germany).

## 2.2. Methods

### 2.2.1.Preparative

The wines were prepared in the facilities of Villa Vinifera (Brestovitsa, Plovdiv, Bulgaria). The grape was pressed in a hydraulic press and 60 mg L<sup>-1</sup> SO<sub>2</sub> and 2g kg<sup>-1</sup> Lallzyme cuvée blanc was added. The must was cooled down to 8°C and when clarification occurred the precipitates were removed by filtration. The filtrated must was transferred to a fermentation vessel, warmed to 15°C and inoculated with LalvinD47  $(0.25 \text{ g } \text{L}^{-1})$ . The must was divided in seven vessels – one control and six variants (17 L each) and to each vessel (without the control one) was added rose waste: W1-8.5g (0.05%); W2-17g (0.1%), W3-42.5g (0.25%), W4-85g (0.5%), W5-170g (1%) and W6-340g (2%). The fermentation continued 22 days at 16±1°C and the solid substances were removed by filtration. A combined agent for wine treatment consisted protein, bentonite of plant and polyvinylpolypyrrolidone (Polymust press) was added (6 g per each). The wines were filtered, bottled with cork stoppers and stored in dark at  $18\pm1^{\circ}C$ .

### 2.2.2.Analytical

The ethanol content was determined by the pycnometric method (Cioch-Skoneczny et al., 2021). Total polyphenols were determined according to Singleton and Rossi (1965) with Folin-Ciocalteu's reagent. Gallic acid was employed as calibration standard and the results were expressed as gallic acid equivalents (GAE) per liter of wine. The antioxidant activities were evaluated by [2,2-diphenyl-1- picrylhydrazyl] Ferric (DPPH) radical and Reducing Antioxidant Power (FRAP) methods as described by Slavov et al. (2017). The amount monomeric anthocyanins of total was determined by the pH-differential method (Giusti and Wrolstad, 2001). Briefly, the wine samples were diluted in parallel with two buffer solutions: 0.025 M KCl with pH 1.0 and 0.4 M sodium acetate with pH 4.5. After one hour at room temperature (22±1°C) absorption at 520 and 700 nm were measured (1 cm cuvette; spectrophotometer Helios Omega UV-Vis with VISIONlite software (Thermo Fisher Scientific, Madison, USA)). The results were calculated using molar absorption coefficient 26900 L mol-<sup>1</sup> cm<sup>-1</sup>, molecular mass of 449.2 g mol<sup>-1</sup> and were expressed as equivalents cyaniding-3-glucoside per liter.

The color characteristics of wines were determined with a Helios Omega UV-Vis spectrophotometer equipped with VISION*lite* ColorCalc Basic software (Thermo Fisher Scientific, USA) using 1 cm cuvettes. Spectra were recorded in a 380–780 nm range at intervals  $\Delta\lambda$ =2 nm. CIELCh color coordinates were calculated using standard illuminant D 65 and 10° observer angle.

The relative proportion of red color from anthocyanes' flavylium cations, dA(%), was calculated using the equation (1), according to Azar *et al.* (1990):

$$dA(\%) = \left(1 - \frac{A420 - A620}{2 \times A520}\right) x \ 100 \quad (1)$$

where, A420, A520 и A620 are the values of absorption at 420, 520 and 620 nm, respectively.

Individual phenolic acids were determined as described by Terzieva et al. (2017) with an HPLC system ELITE LaChrome (Hitachi, Japan) equipped with diode array detector Elite LaChrome L-2455. The separation was performed on Supelco Discovery HS C<sub>18</sub> column (5  $\mu$ m × 25 cm × 4.6 mm) operated at 30°C under gradient conditions with mobile phase consisting of 2% (v/v) acetic acid (mobile phase A) and acetonitrile (mobile phase B) at a flow rate 0.8 mL min<sup>-1</sup>. The gradient used was: 0-1 min: 95% A and 5% B; 1-40 min: 50% A and 50% B; 40-45 min: 100% B; 46-50 min: 95% A and 5% B. The gallic, protocatechuic and cinnamic acids were detected at 280 nm and the chlorogenic, caffeic. ferulic, p-coumaric, sinapic, rosmarinic and chicoric acids - at 320 nm.

The composition of aromatized wines was investigated by gas chromatography with flame ionization detector (GC-FID) and gas chromatography with mass selective detector (GC-MS). The GC-FID analyses were performed on Shimadzu GC-17A (Shimadzu, Japan) equipped with TEKNOKROMA TRB-WAX column  $(30m \times 0.32mm \times 0.25\mu m)$  and software GC Solution (Shimadzu, Japan). Sample amount: 1µL; injector temperature: 229°С; career gas pressure: 32 кРа; career gas speed: 1 mL min<sup>-1</sup>; detector temperature: 250°C; temperature regimen of the column: starting from 40°C, hold for 1 min, increase with 5°C min<sup>-1</sup> until 100°C, hold for 10 minutes and increase with 15°C min<sup>-1</sup> until 220°C.

The GC-MS analyses were performed as follow:

1). Non-volatile polar substances: 0.2 mL ethanolic extract was lyophilized and 50 µL pyridine and 50 µL N,O-Bis-(trimethylsilyl)trifluoroacetamide (BSTFA) were added. The sample was incubated at 70°C for 40 min. For analysis 1.0µL from the solution was injected on gas chromatograph Agilent GC 7890 (Agilent Technologies, Palo Alto, CA, USA) with masselective detector Agilent MD 5975 and column HP-5ms (30m  $\times$ 0.32mm  $\times 0.25 \mu m$ thicknesses). The following temperature regimen was used: initial temperature 100°C (hold for 2 min) then increased to  $180^{\circ}$ C with  $15^{\circ}$ C min<sup>-1</sup> (hold for 1 min) and increase of the temperature to  $300^{\circ}$ C with  $5^{\circ}$ C min<sup>-1</sup> (hold for 10 min); injector and detector temperatures –  $250^{\circ}$ C, helium was used as carrier gas at flow rate 1.0 mL/min. The scanning range of mass-selective detector was m/z = 50 - 550 in split-split mode (10:1).

2). Volatile substances: The aroma substances were extracted according to the procedure described by Uekane et al. (2017). The analyses were performed with gas chromatograph Agilent GC 7890 with masselective detector Agilent MD 5975 and Agilent DB-5ms (30 m  $\times$  0.25 mm  $\times$  0.25 µm) column. The following temperature regimen was used initial temperature was 40°C and then increase to 300°C with 5°C min<sup>-1</sup> (hold for 10 min); injector and detector temperatures - 250°C, helium was used as carrier gas at 1.0mL/min. The scanning range of mass-selective detector was m/z = 40-400 in splitless mode.

The individual compounds were identified comparing the retention times and the relative index (RI) with those of standard substances (linear n-alkanes ( $C_8-C_{40}$ ) injected under the same conditions) and mas-spectral data from libraries of The Golm Metabolome Database (http://csbdb.mpimp-

golm.mpg.de/csbdb/gmd/gmd.html) and NIST'08 (National Institute of Standards and Technology, USA).

#### 2.2.3.Sensory analysis of aromatized wines

Sensory evaluation was performed according to ISO 13299:2016 with the following indicators: color intensity, aroma intensity, fruity nuances, flowery nuances, grassy nuances, taste intensity, acidity, and bitterness. Briefly, the bottles (15±0.5°C) were opened, poured in wineglasses and served coded to 21 (22-52 years old) untrained consumers. The degree of liking was based on an eleven-point scale (0: absence of the specified indicator, 10: extremely sensing the specified parameter). Organoleptic evaluation was done in three repetitions, and the values of individual attributes were averaged and added together.

#### 2.2.4.Statistical analysis

The analyses were performed in triplicate and the data were given as mean values. Statistical significance was detected by analysis of variance (ANOVA, Tukey's HSD test; value of p<0.05 indicated statistical difference).

### 3.Results and discussions

# **3.1.Preparation and characterization of aromatized wines**

## 3.1.1.Preparation and physico-chemical characteristics

Mavrud is among the highly valued local grape varieties. It is specific for the Western Thrace wine region of Bulgaria and traditionally is used for production of red wines. During the last years a tendency for making rosé wines on the basis of Mavrud was observed. The wines are distinctive with well-balanced fruity aroma bouquet and elegant taste of wild forest berries. Six variants with different amounts of added waste and control rosé wine were prepared based on preliminary experiments. At the end of fermentation flavor intensity was amplified in the variants with higher amounts of added rose waste and grassy and incomplete nuances were sensed. After removal of the precipitations and clarification the flavor was significantly improved and more harmonious nuances were detected while the grassy notes significantly decreased. The alcoholic content was in the 14.4-14.6±0.2 % (v) range and pH 3.43-3.62±0.1.

The total monomeric anthocyanins (Table 1) increased in W2 compared to control but then decreased in W4-W6 and the same trend was observed for the relative part of the red color due to flavilium cations of anthocyanins. In general the process of distillation of roses led to destruction of anthocyanins and the waste could not contribute to the final wine anthocyanin content. Slight differences in the color shades were observed: the lower the added rose waste, the more intensive peony color was obtained and this observation was confirmed with the increase of hue angle value –  $46.21\pm0.84$  for the control wine and  $54.95\pm0.70$  for the W6. From the other side the higher content of polyphenols in the

rose waste could probably contribute to stabilization of the anthocyanins in the beverage due to co-pigmentation (Shikov *et al.*, 2012). The total polyphenol content (TPP) increased significantly from  $355.01\pm10.14 \mu$ mol GAE L<sup>-1</sup> for the control wine to  $576.08\pm12.08 \mu$ mol GAE L<sup>-1</sup> for W6 which is due to extraction of

polyphenols during maceration and cofermentation. The increased TPP resulted in significant augmentation of antioxidant activity: for control wine  $668.33\pm21.28$  (by DPPH) and  $918.33\pm15.64$  mg TE L<sup>-1</sup> (by FRAP) compared to W6 - 1991.67±23.95 (by DPPH) and  $2850.00\pm24.85$  mg TE L<sup>-1</sup> (by FRAP).

	С	W1	W2	W3	W4	W5	W6
TMA,	2.62±	2.57±	2.94±	2.52±	2.38±	2.24±	2.06±
mg CG L <sup>-1</sup>	0.10 <sup>a</sup>	0.12ª	0.11 <sup>b</sup>	0.10 <sup>a</sup>	0.11 <sup>a,c</sup>	0.12 <sup>c</sup>	0.11°
TPP,	355.01±	$344.98\pm$	355.27±	375.32±	$395.44\pm$	412.12±	576.08±
µmol GAE L <sup>-1</sup>	10.14 <sup>a,b</sup>	15.21 <sup>a,b</sup>	11.18 <sup>a,b</sup>	12.08 <sup>b,c</sup>	14.51 <sup>c,d</sup>	10.14 <sup>d</sup>	12.08 <sup>e</sup>
DPPH,	$668.33\pm$	558.33±	712.50±	$885.00\pm$	$1016.67\pm$	$1423.33 \pm$	1991.67±
mg TE L <sup>-1</sup>	21.28ª	19.84 <sup>b</sup>	18.30°	21.35 <sup>d</sup>	17.84 <sup>e</sup>	$26.34^{\mathrm{f}}$	23.95 <sup>g</sup>
FRAP,	$918.33\pm$	941.67±	$1085.00 \pm$	$1288.33 \pm$	$1488.33\pm$	$2091.67 \pm$	$2850.00 \pm$
mg TE L <sup>-1</sup>	15.64 <sup>a</sup>	18.41ª	17.74 <sup>b</sup>	16.95°	21.54 <sup>d</sup>	19.62 <sup>e</sup>	$24.85^{\mathrm{f}}$
C (Chrome)	23.10±	22.81±	25.43±	22.81±	21.83±	23.80±	23.94±
C (Chroma)	0.68 <sup>a,c</sup>	0.88 <sup>a,c</sup>	0.72 <sup>b,c</sup>	0.81 <sup>a,c</sup>	0.95ª	0.81 <sup>a,c</sup>	0.72 <sup>c</sup>
h (Huo angla)	46.21±	$45.34\pm$	46.32±	44.33±	46.78±	49.56±	$54.95 \pm$
II (IIUe angle)	0.84ª	1.01 <sup>a</sup>	0.74 <sup>a</sup>	0.83ª	0.84ª	0.68 <sup>b</sup>	0.70 <sup>c</sup>
I (Lightness)	82.10±	$77.07\pm$	79.96±	77.88±	76.06±	$80.34\pm$	79.86±
L (Lightness)	0.88ª	1.15 <sup>b,c</sup>	1.08 <sup>b,a</sup>	1.17 <sup>b,c</sup>	1.21°	1.35 <sup>b,a</sup>	1.41 <sup>b,a</sup>
9	17.22±	17.11±	$18.84\pm$	17.25±	15.74±	16.42±	$14.78\pm$
a	0.45 <sup>a</sup>	0.61ª	0.48 <sup>b</sup>	0.51ª	0.49°	0.54 <sup>a,c</sup>	0.57 <sup>d</sup>
h	15.51±	15.12±	17.17±	15.04±	15.12±	17.37±	18.92±
U	0.41 <sup>a</sup>	0.61 <sup>a</sup>	0.48 <sup>b</sup>	0.74 <sup>a</sup>	0.46 <sup>a</sup>	0.50 <sup>b</sup>	0.49 <sup>c</sup>
	62.53±	65.28±	53.32±	52.15±	56.21±	46.34±	41.02±
<b>CI, <i>U</i>A %</b>	1.05ª	1.16 <sup>b</sup>	1.21°	1.30°	1.50 <sup>d</sup>	1.17 <sup>e</sup>	1.24 <sup>f</sup>

 Table 1. Physico-chemical characteristics of control and aromatized wines

TMA – total monomeric anthocyanins; CG – cyanind-3-glucoside; TPP – total polyphenolic content; GAE – gallic acid equivalents; TE – Trolox equivalents; CI, dA% – relative part of the red color due to flavilium cations of anthocyanins; <sup>a, b, c, d, e, f, g</sup> Values with different letters in a raw are statistically different (Tukey's HSD test, p < 0.05)

#### 3.1.2.HPLC determination of phenolic acids

Individual phenolic acids were determined by HPLC (Table 2). The highest increase was detected for gallic acid and 3,4-dihydroxy benzoic acid: from  $1.2\pm0.9 \text{ mg L}^{-1}$  and  $5.4\pm0.2 \text{ mg L}^{-1}$  in the control wine to  $25.9\pm0.9 \text{ mg L}^{-1}$ and  $65.1\pm1.1 \text{ mg L}^{-1}$  in W6, respectively. The increase of phenolic acids quantity in W3-W6 compared to control and W1-W2 could be explained with the addition of higher amounts of rose waste and subsequent extraction during fermentation, having in mind that the rose wastes are rich source of polyphenols (*Shikov et al.*, 2012). The higher amounts of phenolic acids determined was also related to increase in the antioxidant capacity of the aromatized wines from W1 to W6 (Table 1) and this could be explained with the higher amounts of total polyphenols extracted from the rose waste but not the anthocyanins (no significant difference in the TMA amounts in all wines).

Compound, mg L <sup>-1</sup>	С	W1	W2	W3	W4	W5	W6
Gallic acid	1.2±0.9 <sup>a</sup>	3.3±0.9 <sup>b</sup>	$4.1 \pm 0.8^{b}$	8.8±0.7°	17.9±0.5 <sup>d</sup>	$19.2{\pm}0.7^{d}$	25.9±0.9e
3,4-dihydroxy benzoic acid	5.4±0.2 <sup>a</sup>	9.8±0.4 <sup>b</sup>	15.0±0.6°	22.4±0.3 <sup>d</sup>	$24.2{\pm}0.9^{d}$	32.4±1.0 <sup>e</sup>	$65.1 \pm 1.1^{f}$
Chlorogenic acid	traces	0.1±0.0ª	traces	traces	10.8±0.5 <sup>b</sup>	11.7±0.6 <sup>b</sup>	10.2±0.8 <sup>b</sup>
Caffeic acid	$0.7{\pm}0.2^{a}$	$0.7{\pm}0.2^{a}$	$0.7{\pm}0.2^{a}$	1.8±0.1 <sup>b</sup>	1.9±0.1 <sup>b</sup>	2.3±0.2 <sup>b</sup>	3.4±0.2°
Ferulic acid	traces	nd	traces	$0.1 \pm 0.0^{a}$	traces	traces	$0.2{\pm}0.0^{a}$
p-Coumaric acid	2.7±0.5ª	2.7±0.5ª	2.7±0.5ª	2.5±0.5ª	3.2±0.4 <sup>a,b</sup>	4.0±0.7 <sup>b,c</sup>	4.5±0.3°
Sinapic acid	1.0±0.3ª	$0.9{\pm}0.3^{a}$	1.0±0.2ª	1.1±0.4 <sup>a</sup>	1.2±0.2 <sup>a</sup>	1.5±0.3 <sup>a,b</sup>	$1.8 \pm 0.3^{b}$
Rosmarinic acid	traces	traces	traces	traces	traces	traces	0.9±0.2
Cinnamic acid	traces	traces	traces	traces	traces	$0.1{\pm}0.0^{a}$	$0.2{\pm}0.0^{a}$

**Table 2.** Phenolic acids in control and aromatized wines

<sup>a, b, c, d,e</sup> Values with different letters in a raw are statistically different (Tukey's HSD test, p < 0.05)

 Table 3. GC-FID analysis of control and aromatized wines

Compound, mg L <sup>-1</sup>	С	W1	W2	W3	W4	W5	W6
Aastaldahyda	17.1±	19.9±	20.3±	22.2±	24.1±	25.2±	27.0±
Acetaiuenyue	1.2ª	0.9ª	1.3 <sup>a,b</sup>	1.0 <sup>b,c</sup>	0.9°	1.1 <sup>c,d</sup>	0.9 <sup>d</sup>
Ethyl agotata	32.7±	33.8±	34.7±	33.3±	36.0±	37.6±	37.7±
Ethyl acetate	2.1ª	2.3ª	2.0ª	2.3ª	1.9 <sup>a,b</sup>	1.8 <sup>b</sup>	1.9 <sup>b</sup>
Mathanal	34.2±	35.1±	35.2±	36.4±	37.2±	38.7±	39.1±
Methanoi	2.1ª	2.0ª	2.3ª	2.1 <sup>a,b</sup>	2.5 <sup>a,b</sup>	2.6 <sup>b</sup>	1.8 <sup>b</sup>
2 hutanal	29.4±	30.2±	32.8±	33.3±	29.0±	33.7±	34.0±
2-butanoi	0.9ª	1.2 <sup>a,b</sup>	1.2 <sup>a,b</sup>	1.1 <sup>b</sup>	1.1 <sup>a</sup>	1.2 <sup>b</sup>	1.2 <sup>b</sup>
1 nronanal	45.8±	45.9±	44.9±	46.4±	46.1±	47.7±	48.0±
1-propanoi	0.9ª	1.0 <sup>a</sup>	1.1 <sup>a</sup>	1.0 <sup>a,b</sup>	1.1 <sup>a,b</sup>	1.1 <sup>a,b</sup>	1.0 <sup>b</sup>
i hutanal	27.3±	24.2±	28.1±	23.8±	25.6±	26.2±	$28.3\pm$
I-Dutanoi	1.5 <sup>a,b</sup>	1.4 <sup>b,c</sup>	$1.6^{a,b,d}$	1.5°	1.7 <sup>a,b,c,d</sup>	1.7 <sup>a,b,c,d</sup>	1.3 <sup>d</sup>
1 hutanal	7.0±	7.2±	$8.0\pm$	8.6±	8.7±	9.0±	9.8±
1-Dutanoi	0.3ª	0.3ª	0.2 <sup>b</sup>	0.4 <sup>b,c</sup>	0.2 <sup>b,c</sup>	0.3 <sup>c,d</sup>	0.3 <sup>d</sup>
i-amyl	7.9±	$8.7\pm$	$8.7\pm$	$8.6\pm$	8.6±	$8.7\pm$	$8.9\pm$
alcohols	$0.4^{a}$	0.5ª	0.4ª	0.3ª	0.4ª	0.5ª	0.4ª
Sum of higher	117.4±	$116.2 \pm$	122.5±	$120.7\pm$	$118.0\pm$	125.3±	129.0±
alcohols	1.5ª	1.4ª	1.6 <sup>b</sup>	1.5 <sup>a,b</sup>	1.7ª	1.7 <sup>b</sup>	1.3°

<sup>a, b, c, d</sup> Values with different letters in a raw are statistically different (Tukey's HSD test, p < 0.05)

### 3.1.3.GC-FID analyses

The GC-FID analysis (Table 3) revealed slight increase of acetaldehyde, ethyl acetate and methanol from W1 to W6 compared to control. The increased methanol content could be explained with the presence of pectic substance in the rose waste (Slavov et al., 2017). Nevertheless, the amounts determined (even in W6) were within the permissible limits: for example the methanol limit is  $250 \text{ mg } \text{L}^{-1}$  for white and rosé wines and the higher amount observed in W6 was 39.1±1.8 mg L<sup>-1</sup> (Compendium of international methods of analysis - OIV, 2018). The quantity of ethyl acetate determined was in the  $32.7\pm2.1$  –  $37.7\pm1.9$  mg L<sup>-1</sup> range. An aroma similar to acetone is sensed if the concentration of ethyl acetate exceeds a threshold reported most often as being between 100-200 mg L<sup>-1</sup> (Cliff and Pickering, 2006). Ethyl acetate concentrations below the threshold can contribute to the depth of body, richness and sweetness of wine and between 30-80 mg L<sup>-1</sup> ethyl acetate can add to the wine character and be a part of the pleasant wine bouquet (Plata et al., 2003). The amounts of higher alcohols increased in the W1-W6 variants (except for i-amyl alcohols) compared to the control wine. Higher alcohols, also known

as fusels, plays and important role in the formation of wine aroma and at concentrations below 300 mg L<sup>-1</sup> positively influence aroma. The higher amounts negatively affect the proper bouquet of the wine. The aromatized wines and the control rosé had a total amounts of fusels in the 117.4-129.0 $\pm$ 1.5 mg L<sup>-1</sup> range and it could be concluded that addition of rose waste during fermentation did not affect negatively formation of higher alcohols.

## 3.1.4.GC-MS analyses – determination of polar volatile and non-volatile compounds

Furthermore the aromatized wines were subjected to GC-MS analysis. The preparation of rosé wines with Mavrud grape is a rare practice since this regional grape variety is mostly used for red wine preparation and to the best of our knowledge this is the first report for GC-MS profiling of Mavrud rosé. As a result of the analysis thirty nine polar non-volatile metabolites (amino acids, sugars, acids, sugar alcohols and sterols) were tentatively detected in the control and aromatized wines. In general increase in the content of most of the detected substances from control to W6 was observed (Table 4 and 5).

Compound	RI	С	W1	W2	W3	W4	W5	W6				
Compound	M		% of TIC									
Lactic acid	1066	25.1±	24.2±	24.8±	26.1±	24.3±	27.5±	29.2±				
Lattit atlu		0.7 <sup>a, b</sup>	0.6 <sup>a</sup>	0.8 ª	0.5 <sup>b</sup>	0.9 <sup>a,b</sup>	0.4 <sup>b,c</sup>	0.6 °				
L-Valine	1778	10.6±	11.4±	12.3±	13.2±	13.9±	15.2±	15.9±				
	1220	0.8 <sup>a</sup>	0.9 <sup>a, b</sup>	0.8 <sup>a, b, c</sup>	0.7 <sup>b,c,d</sup>	0.9 <sup>c,d,e</sup>	0.7 <sup>d,e</sup>	0.9 <sup>e</sup>				
Character	1266	$426.3\pm$	$398.8\pm$	431.7±	532.9±	$586.8\pm$	576.7±	592.4±				
Glycerol		1.6 <sup>a</sup>	1.4 <sup>b</sup>	1.6 °	2.1 <sup>d</sup>	2.5 <sup>e</sup>	2.0 <sup>f</sup>	1.8 <sup>e</sup>				
I Louging	1777	11.0±	11.9±	12.7±	13.8±	14.5±	12.9±	15.2±				
L-Leucine	1272	0.7 <sup>a</sup>	0.8 <sup>a, b</sup>	0.9 <sup> a, b, c</sup>	0.6 <sup>b,c</sup>	0.7 °	0.9 <sup>a,b,c</sup>	0.8 °				
Dhaspharia agid	1770	88.0±	85.2±	90.3±	92.9±	94.2±	93.3±	95.4±				
r nosphoric aciu	1270	1.0 <sup>a,b</sup>	0.9 <sup>a</sup>	1.2 <sup>b</sup>	1.1 <sup>b,c</sup>	1.0 °	1.5 <sup>b,c</sup>	1.2 °				
I Isolouaire	1200	10.1±	10.8±	11.4±	12.6±	14.3±	14.9±	15.6±				
L-isoleucine	1299	0.5 <sup>a</sup>	0.6 <sup>a</sup>	0.5 <sup>a, b</sup>	0.6 <sup>b</sup>	0.7 °	0.5 °	0.7 °				
I Dualina	1207	42.0±	41.5±	49.3±	52.5±	58.4±	57.5±	59.9±				
L-Proline	1307	0.9 <sup>a</sup>	0.7 <sup>a</sup>	0.8 <sup>b</sup>	1.0 °	0.9 <sup>d</sup>	$0.8^{d}$	0.9 <sup>d</sup>				

**Table 4.** Polar non-volatile metabolites in control and aromatized wines

~		188.2±	187.2±	190.1±	191.2±	195.4±	196.8±	195.1±
Succinic acid	1310	1.9 <sup>a</sup>	2.5 <sup>a</sup>	2.0 <sup>a, b</sup>	1.8 <sup>a,b</sup>	1.9 <sup>b</sup>	1.6 <sup>b</sup>	1.7 <sup>b</sup>
	1220	55.7±	57.6±	62.4±	69.7±	72.9±	74.0±	75.4±
Giyceric acid	1559	1.8 <sup>a</sup>	1.2 ª	1.0 <sup>b</sup>	1.4 °	1.0 <sup>d</sup>	1.1 <sup>d,e</sup>	1.2 °
Eumonia agid	1255	28.0±	27.9±	30.3±	35.0±	36.9±	38.3±	39.9±
Fulliaric acid	1555	1.1 ª	0.9 ª	0.8 <sup>a</sup>	1.0 <sup>b</sup>	0.8 <sup>b,c</sup>	0.9 <sup>c,d</sup>	1.0 <sup>d</sup>
Sorino	1367	12.4±	$13.5\pm$	12.4±	15.5±	16.8±	16.0±	$17.1\pm$
Serme	1302	0.6 <sup>a</sup>	0.4 <sup>a</sup>	0.8 <sup>a</sup>	0.5 <sup>b,d</sup>	0.4 <sup>c, d</sup>	0.5 <sup>d</sup>	0.6 <sup>c,d</sup>
I -Threonine	1390	14.8±	16.0±	17.3±	18.4±	20.3±	21.4±	$22.5\pm$
	1590	0.8 <sup>a</sup>	0.7 <sup>a</sup>	0.5 <sup>b</sup>	0.4 °	0.7 <sup>d</sup>	0.5 <sup>d,e</sup>	0.6 °
L-Malic acid	1488	297.0±	298.8±	299.9±	301.3±	298.4±	302.3±	305.6±
		2.0 <sup>a</sup>	1.1 ª	1.0 <sup>a</sup>	1.9 <sup>a,b</sup>	1.8 <sup>a</sup>	1.9 <sup>a,b</sup>	1.8 °
Pyroglutamic	1512	65.6±	69.8±	74.2±	82.0±	85.8±	$86.7\pm$	$88.3\pm$
acid		0.9ª	$0.8^{\circ}$	$0.7^{\circ}$	$0.6^{\rm u}$	0.8	$0.8^{\text{c,i}}$	0.91
Salycilic acid	1516	$24.0\pm$	23.9±	$26.3\pm$	$2/.0\pm$	23.2±	26.9±	23.8±
		0.4 *	0.8 <sup>-0,0</sup>	12.8	$0.8^{\circ}$	$0.7^{a,c}$	0.3°	0.4 <sup>4,0</sup>
L-Aspartic acid	1531	$11.0\pm$	$12.1\pm$	$12.0\pm$ 0 Q a, b, c	$15./\pm$ 0 0b.c.d	$14.1\pm$ 0.7c,d	$13.2\pm$	$13.9\pm$
		0.8 88 0+	0.9 05 2±	$104.3 \pm$	110.9	$100.7^{+}$	111 / -	$112.0\pm$
L-Threonic acid	1528	00.0± 1.2ª	95.2± 10 <sup>b</sup>	104.5±	1 3 d	$109.2\pm$	111.4±	112.9±
L_(+)_Tartaric		$\frac{1.2}{224.3+}$	278.4+	222 1+	230.8+	231.1	229.9+	235.0+
	1612	$1.6^{a}$	17 <sup>b</sup>	$1 4^{a}$	15 <sup>b</sup>	1 1 <sup>b</sup>	1 2 <sup>b</sup>	13°
		13.4±	$14.4\pm$	$15.7\pm$	$16.8\pm$	$17.3\pm$	$17.8\pm$	$18.0\pm$
L-Phenylalanine	1646	0.9 ª	0.8 <sup>a, b</sup>	$0.7^{a, b, c}$	1.0 <sup>b,c,d</sup>	0.8 <sup>c,d</sup>	$0.7^{\rm d}$	$0.5^{d}$
** ••••	1750	16.1±	16.9±	19.5±	20.2±	21.1±	22.0±	22.8±
Vanillic acid	1758	1.0 ª	0.7 <sup>a</sup>	0.5 <sup>b</sup>	0.7 <sup>b,c</sup>	0.6 <sup>c,d</sup>	0.7 <sup>d</sup>	0.9 <sup>d</sup>
Protocatechuic	1012	18.6±	19.5±	21.2±	23.3±	24.9±	25.3±	26.1±
acid	1815	0.6 <sup>a</sup>	0.5 <sup>a</sup>	0.8 <sup>b</sup>	0.5 °	0.8 <sup>d</sup>	0.9 <sup>d</sup>	0.5 <sup>d</sup>
Quinic acid	18/13	22.5±	24.3±	27.8±	28.1±	29.8±	$30.5\pm$	30.4±
Quint actu	10+3	0.8 <sup>a</sup>	0.9 <sup>a</sup>	0.8 <sup>b</sup>	0.7 <sup>b</sup>	0.9 <sup>b,c,d</sup>	0.6 <sup>c,d</sup>	0.7 <sup>d</sup>
Fructose	1862	$66.5\pm$	70.1±	$78.3\pm$	83.1±	85.0±	$88.2\pm$	$90.7\pm$
1140050	1002	1.4 <sup>a</sup>	0.9 <sup>b</sup>	1.0 °	1.3 <sup>d</sup>	1.0 d	1.1 <sup>e</sup>	0.9 <sup>e</sup>
Galactose	1884	62.3±	66.7±	69.2±	77.9±	79.3±	80.4±	$80.8\pm$
		$0.8^{a}$	0.9 °	1.1°	0.9 <sup> a</sup>	0.8 <sup>d,e</sup>	0.9 <sup>a,e</sup>	0.8 °
Syringic acid	1888	$16.1\pm$	17.2±	19.5±	$20.2\pm$	19.1±	$20.5\pm$	21.4±
		$0.6^{a}$	$0.7^{a}$	0.8 0	$0.8^{0, c}$	$0.7^{\circ}$	$0.6^{0,c}$	$0.7^{\circ}$
Glucose	1896	169.0±	$181.5\pm$	$208.4\pm$	$211.3\pm$	219.4±	$215.8\pm$	$218./\pm$
		1.9"	1.4	1.5	1.8°	1.9 "	1.4°	1.5°,°
Glucitol	1930	00.9±	03.4± 07 <sup>b</sup>	00.9±	/0.1±	//.ð±	/3.9±	/0.3± 0.8d
		$36.8 \pm$	0.7 37 0⊥	$\frac{0.3}{41.2\pm}$	$46.0\pm$	<u> </u>	0.9 46 0±	0.0 17 &⊥
Gluconic acid	1991	$0.0\pm$	$0.8^{\pm}$	$1.2\pm$ 0.7 <sup>b</sup>	10.0± 0.7°	13.0± 0.8°	1 ∩°	+7.0± 11°
		54 7+	59 8+	63.5+	68 4+	66 3+	67.5+	67.6+
Palmitic acid	2039	$1.0^{a}$	1.1 <sup>b</sup>	1.0°	0.9 <sup>d</sup>	$0.8^{d}$	0.9 <sup>d</sup>	$0.7^{\rm d}$
		27.7±	$28.5\pm$	30.8±	34.6±	35.7±	33.9±	36.8±
Glucaric acid	2013	0.7 <sup>a</sup>	0.8 <sup>a, b</sup>	0.8 <sup>b</sup>	0.9 °	1.0 °	1.1 °	1.0°

Myo Inosital	2000	$8.4\pm$	$8.9\pm$	9.2±	$10.5\pm$	10.9±	9.9±	$10.8\pm$
wiyo-inositoi	2090	0.6 <sup>a</sup>	0.5 <sup>a, b</sup>	0.6 <sup>a, b, c</sup>	$0.7^{b,c}$	0.5 °	$0.8^{a,b,c}$	0.6 <sup>b,c</sup>
Steenie eeid	2122	32.7±	31.5±	35.7±	40.9±	42.4±	43.1±	43.8±
Stearic acid	2132	1.1 <sup>a</sup>	1.2 ª	1.3 <sup>b</sup>	1.0 °	0.8 <sup>c,d</sup>	0.9 <sup>d</sup>	0.9 <sup>d</sup>
	2140	14.5±	15.2±	16.9±	18.2±	18.9±	18.5±	19.3±
Carreic acid	2140	0.9 <sup>a</sup>	0.7 <sup>a</sup>	0.8 <sup>a, b</sup>	$0.8^{b,c}$	0.5 °	0.7 °	0.6 °
Linoleic acid	2200	48.3±	51.9±	55.5±	60.3±	62.5±	61.9±	64.1±
	2209	1.0 <sup>a</sup>	$0.8^{b}$	0.9 °	1.1 <sup>d</sup>	1.0 <sup>d,e</sup>	1.1 <sup>d,e</sup>	1.0 <sup>e</sup>
	2217	23.9±	24.8±	28.1±	29.8±	31.2±	32.4±	30.3±
a-Linolenic acid		0.8 <sup>a</sup>	0.7 <sup>a</sup>	0.4 <sup>b</sup>	0.9 °	0.8 <sup>c,d</sup>	$0.7^{\rm d}$	0.8 <sup>c,d</sup>
Success	2(40	63.4±	$66.8\pm$	68.9±	79.2±	77.8±	78.9±	79.9±
Sucrose	2649	1.1 <sup>a</sup>	1.0 <sup>b</sup>	1.1 <sup>b</sup>	1.0 °	1.2 °	1.1 °	1.0 °
Tunanaga	2742	35.1±	39.8±	42.7±	43.9±	42.3±	41.9±	40.8±
Iuranose	2742	1.3 <sup>a</sup>	1.0 <sup>b</sup>	1.2 °	1.1 °	1.2 °	1.0 <sup>b,c</sup>	1.1 <sup>b,c</sup>
Stigmagtanal	2215	12.3±	14.1±	13.3±	15.4±	16.0±	15.8±	16.7±
Sugmasterol	5515	0.7 <sup>a</sup>	0.8 <sup>a, b</sup>	0.9 <sup> a, b</sup>	0.8 <sup>b,c</sup>	0.7 °	0.9 °	0.8 °
R Sitestans	2255	11.7±	12.9±	12.7±	14.6±	15.7±	15.4±	15.8±
p-Situsterol	2222	0.8 <sup>a</sup>	0.7 <sup>a, b</sup>	0.5 <sup>a</sup>	0.7 <sup>b,c</sup>	0.8 °	0.7 °	0.8 °

RI: relative index (Kovats retention index)

TIC: total ion current

The results are presented as mean  $\pm$  SD (n=3) <sup>a, b, c, d, e, f, g</sup> Values with different letters in a raw are statistically different (Tukey's HSD test, p < 0.05)

Table 5. P	olar volatile	(aroma)	) substar	nces in c	ontrol and	aromatize	ed wines

Compound	RI	С	W1	W2	W3	W4	W5	W6				
			% of TIC									
Alcohols												
Propan_1_ol	599	0.36±	0.35±	$0.36\pm$	0.37±	0.38±	0.38±	0.39±				
1 Topan-1-01	577	0.08 <sup>a</sup>	0.05ª	0.06 <sup>a</sup>	0.07 <sup>a</sup>	0.05 <sup>a</sup>	0.04 <sup>a</sup>	0.08 <sup>a</sup>				
Buton 1 ol	660	$0.30\pm$	$0.30\pm$	0.31±	$0.31\pm$	$0.30\pm$	$0.30\pm$	$0.30\pm$				
Dutail-1-01	000	0.08 <sup>a</sup>	0.07 <sup>a</sup>	0.04 <sup>a</sup>	0.05 <sup>a</sup>	0.06 <sup>a</sup>	0.06 <sup>a</sup>	0.05 <sup>a</sup>				
Donton 1 ol	768	$0.40\pm$	$0.41\pm$	$0.40\pm$	$0.40\pm$	$0.41\pm$	$0.42\pm$	$0.42\pm$				
I entan-1-01		0.04 <sup>a</sup>	0.02 <sup>a</sup>	0.04 <sup>a</sup>	0.05 <sup>a</sup>	0.03 <sup>a</sup>	0.04 <sup>a</sup>	0.02 <sup>a</sup>				
Hoven 1 ol	867	$1.80\pm$	1.79±	$1.84\pm$	$1.85\pm$	1.86±	1.86±	$1.87\pm$				
1102411-1-01		0.10 <sup>a</sup>	0.10 <sup>a</sup>	0.09 <sup>a</sup>	0.09 <sup>a</sup>	$0.07^{a}$	$0.08^{a}$	0.06 <sup>a</sup>				
Honton 1 ol	012	$0.48\pm$	$0.49\pm$	$0.49\pm$	$0.50\pm$	$0.50\pm$	$0.50\pm$	$0.50\pm$				
neptali-1-01	912	0.07 <sup>a</sup>	0.06 <sup>a</sup>	0.05 <sup>a</sup>	0.05 <sup>a</sup>	$0.07^{a}$	0.06 <sup>a</sup>	0.05 <sup>a</sup>				
Octor 1 ol	002	$0.60\pm$	$0.59\pm$	0.61±	0.61±	0.61±	0.61±	$0.62\pm$				
Octail-1-01	995	0.06 <sup>a</sup>	0.06 <sup>a</sup>	$0.07^{a}$	0.06 <sup>a</sup>	0.06 <sup>a</sup>	0.07 <sup>a</sup>	0.06 <sup>a</sup>				
Nonan 1 al	1170	$0.21\pm$	0.21±	$0.21\pm$	$0.21\pm$	$0.21\pm$	$0.22\pm$	$0.23\pm$				
Inollall-1-01	1170	0.01 <sup>a</sup>	0.01 <sup>a</sup>	$0.01^{a}$	$0.02^{a}$	0.03 <sup>a</sup>	$0.02^{a}$	0.03 <sup>a</sup>				
Decen 1 ol	1272	0.25±	0.26±	$0.27\pm$	0.26±	$0.27\pm$	$0.27\pm$	0.28±				
Decali-1-01	1272	0.04 <sup>a</sup>	0.03 <sup>a</sup>	0.04 <sup>a</sup>	0.03 <sup>a</sup>	0.04 <sup>a</sup>	0.04 <sup>a</sup>	0.06 <sup>a</sup>				
Acids			-									
A cotio soid	640	0.41±	0.41±	$0.42\pm$	0.42±	$0.41\pm$	$0.42\pm$	$0.43\pm$				
Acenc aciu	040	0.06 <sup>a</sup>	0.05 <sup>a</sup>	$0.06^{a}$	0.07 <sup>a</sup>	0.05 <sup>a</sup>	0.06 <sup>a</sup>	0.04 <sup>a</sup>				

Butanoic acid	785	0.32±	0.31±	0.31±	0.33±	0.33±	0.33±	0.33±
Dutanoit atra	105	0.07 <sup>a</sup>	0.06 <sup>a</sup>	0.05 <sup>a</sup>	0.04 <sup>a</sup>	0.04 <sup>a</sup>	0.05 <sup>a</sup>	0.06 <sup>a</sup>
Octanoic acid	1192	5.88±	5.89±	5.95±	6.07±	6.08±	6.11±	6.13±
		0.09 <sup>a</sup>	$0.08^{a}$	0.06 <sup>a</sup>	0.09 <sup>a</sup>	$0.08^{a}$	$0.05^{a}$	0.04 <sup>a</sup>
Nonanoic acid	1281	0.76±	$0.77\pm$	$0.78\pm$	$0.78\pm$	0.79±	0.78±	0.79±
		0.10 <sup>a</sup>	0.08 <sup>a</sup>	0.06 <sup>a</sup>	0.09 <sup>a</sup>	$0.07^{a}$	$0.08^{a}$	$0.07^{a}$
Decanoic acid	1388	6.90±	6.95±	7.03±	7.12±	7.09±	7.13±	7.15±
		0.11 ª	$0.08^{a}$	0.09 <sup>a</sup>	0.12 <sup>a</sup>	0.09 <sup>a</sup>	0.06 <sup>a</sup>	0.05ª
Dodecanoic acid	1573	1.48±	1.49±	1.50±	$1.53\pm$	1.54±	$1.55\pm$	1.56±
2000000000	10,0	0.08 ª	0.05ª	0.06 <sup>a</sup>	0.09 <sup>a</sup>	0.08 <sup>a</sup>	0.07ª	0.05 <sup>a</sup>
Tetradecanoic acid	1774	$0.30\pm$	$0.30\pm$	$0.31\pm$	$0.31\pm$	0.32±	$0.32\pm$	$0.33\pm$
		0.06 ª	$0.04^{a}$	0.03ª	0.04ª	0.05ª	$0.04^{a}$	0.03ª
Aldehydes								
		0.27	0.201	0.201	0.281	0.24	0.281	0.42
Acetaldehyde	400	$0.2/\pm$	$0.30\pm$	$0.30\pm$	$0.28\pm$	$0.34\pm$	$0.38\pm$	$0.42\pm$
		0.00	0.05	1.74	0.03	1.79	0.00	0.03
Hexanal	800	$1.72\pm$ 0.00 a	$1.75\pm$ 0.05 <sup>a</sup>	$1./4\pm$	$1.//\pm$ 0.10 <sup>a</sup>	$1.70^{\pm}$	$1.79\pm$ 0.07 <sup>a</sup>	$1.70\pm$
		0.09	0.03	0.08	0.10	0.09	0.07	0.08
Decanal	1205	$0.0/\pm$	$0.07^{\pm}$	$0.00\pm$	$0.90\pm$	$0.90\pm$ 0.07 <sup>a</sup>	$0.91\pm$	$0.92\pm$
		0.08	0.07	0.00	0.09	0.07	0.00	0.07
Hydrocarbons			1		1	1	1	1
Hexadecane	1600	2.81±	$2.83\pm$	$2.86\pm$	2.91±	2.92±	2.93±	$2.95\pm$
IICAuuccunc	1000	0.10 <sup>a</sup>	0.09 <sup>a</sup>	$0.08^{a}$	0.11 <sup>a</sup>	$0.08^{a}$	0.09 <sup>a</sup>	0.10 <sup>a</sup>
Octadecane	1800	3.34±	3.36±	3.41±	3.45±	3.44±	3.46±	3.48±
	1000	0.11 <sup>a</sup>	$0.08^{a}$	0.09 <sup>a</sup>	0.12 <sup>a</sup>	$0.09^{a}$	0.10 <sup>a</sup>	0.09 <sup>a</sup>
Nonadecane	1900	2.68±	2.69±	2.71±	2.76±	2.77±	2.75±	2.79±
		0.09 <sup>a</sup>	$0.07^{a}$	0.08 <sup>a</sup>	0.14 <sup>a</sup>	$0.08^{a}$	0.10 <sup>a</sup>	$0.08^{a}$
Eicosane	2000	$2.01\pm$	2.02±	$2.05\pm$	2.08±	2.08±	2.10±	2.12±
		0.12 ª	$0.08^{a}$	0.06 <sup>a</sup>	0.09 <sup>a</sup>	0.09 <sup>a</sup>	$0.08^{a}$	$0.07^{a}$
Heneicosane	2100	2.10±	2.11±	2.15±	2.17±	2.18±	2.19±	2.22±
		$0.10^{a}$	$0.08^{a}$	0.07 <sup>a</sup>	0.09 <sup>a</sup>	0.08 <sup>a</sup>	0.09 <sup>a</sup>	$0.10^{a}$
Docosane	2200	1.76±	1.78±	1.77±	1.82±	$1.85\pm$	1.87±	1.89±
		0.08 "	0.06"	0.09"	0.11"	0.10"	0.08"	0.11 <sup>ª</sup>
Terpenes								
Linglool	1097	1.53±	1.54±	$1.57\pm$	1.58±	1.63±	1.69±	1.75±
	1097	0.10 <sup>a</sup>	0.09 <sup>a</sup>	$0.08^{a}$	0.14 <sup>a</sup>	0.09 <sup>a</sup>	0.09 <sup>a</sup>	$0.08^{a}$
Dhonothyl clashel	1110		$1.48\pm$	$1.95\pm$	2.18±	2.68±	2.99±	3.44±
Phenetnyi alconol	1110	na	0.11 <sup>a</sup>	$0.08^{b}$	0.14 <sup>b</sup>	0.11 <sup>c</sup>	0.10 <sup>d</sup>	0.12 <sup>e</sup>
		0.08±	0.49±	0.57±	0.69±	0.86±	1.24±	1.68±
Cis-Rose oxide	1112	0.05 <sup>a</sup>	0.11 <sup>b</sup>	$0.10^{b}$	0.13 <sup>b,c</sup>	0.14 <sup>c</sup>	0.09 <sup>d</sup>	0.08 <sup>e</sup>
		0.10±	0.57±	0.61±	0.59±	0.89±	1.11±	1.79±
Trans-Rose oxide	1127	0.06 <sup>a</sup>	0.05 <sup>b</sup>	0.08 <sup>b</sup>	0.07 <sup>b</sup>	0.11 <sup>b</sup>	0.09 <sup>c</sup>	0.08 <sup>d</sup>
	1000	1	0.12±	0.15±	0.16±	0.18±	0.19±	0.25±
p-Citronellol	1228	nd	$0.02^{a}$	0.03 <sup>a,b</sup>	0.01 <sup>a,b</sup>	0.02 <sup>b</sup>	0.01 <sup>b</sup>	0.02 <sup>c</sup>
C	1055	1	0.63±	0.69±	0.78±	1.25±	1.68±	2.19±
Geraniol	1255	nd	0.11 <sup>a</sup>	$0.10^{a}$	0.09 <sup>a</sup>	0.11 <sup>b</sup>	0.12 <sup>c</sup>	0.14 <sup>d</sup>
Fugaral	1256	2.10±	2.15±	2.18±	2.17±	2.19±	2.22±	2.28±
Eugenoi	1550	0.09 <sup>a</sup>	0.10 <sup>a</sup>	$0.07^{a}$	$0.08^{a}$	0.09 <sup>a</sup>	$0.08^{a}$	$0.08^{a}$
0 Dourhonous	1202	1.26±	1.27±	1.29±	1.30±	1.33±	1.34±	1.38±
<i>p</i> -bourdonene	1383	0.11 <sup>a</sup>	$0.09^{a}$	$0.08^{a}$	$0.10^{a}$	$0.07^{a}$	$0.06^{a}$	$0.05^{a}$

l Flomono	1200	2.54±	2.57±	2.63±	2.62±	2.64±	2.66±	2.69±
<i>p</i> -Elemene	1390	0.12 <sup>a</sup>	0.09 <sup>a</sup>	$0.10^{a}$	0.08 <sup>a</sup>	0.06 <sup>a</sup>	0.05 <sup>a</sup>	$0.04^{a}$
β-Caryophyllene	1410	2.93±	2.95±	2.98±	3.02±	3.03±	3.08±	3.10±
	1419	0.06 <sup>a</sup>	$0.07^{a,b}$	$0.08^{a,b}$	0.09 <sup>a,b</sup>	$0.08^{a,b}$	$0.06^{a,b}$	$0.07^{b}$
l Cubabana	1290	0.76±	0.76±	$0.77\pm$	0.78±	$0.80\pm$	0.85±	0.89±
<i>p</i> -Cubebene	1369	0.05 <sup>a</sup>	$0.07^{a}$	$0.08^{a}$	0.10 <sup>a</sup>	0.05 <sup>a</sup>	$0.04^{a}$	0.03 <sup>a</sup>
α-Guaiene	1/29	$1.08\pm$	$1.08\pm$	1.09±	1.11±	1.15±	1.18±	1.22±
	1430	0.11 <sup>a</sup>	0.10 <sup>a</sup>	0.06 <sup>a</sup>	0.10 <sup>a</sup>	0.12 <sup>a</sup>	$0.07^{a}$	$0.06^{a}$
п	1454	1.51±	1.53±	1.55±	1.56±	1.55±	1.59±	1.67±
a-numulene		0.09 <sup>a</sup>	$0.08^{a}$	$0.08^{a}$	$0.08^{a}$	$0.07^{a}$	$0.08^{a}$	$0.07^{a}$
(7) & Farmasana	1450	1.81±	1.82±	1.86±	1.87±	1.96±	2.05±	2.15±
$(\mathbf{Z})$ -p-rarnesene	1439	0.08 <sup>a</sup>	0.09 <sup>a</sup>	$0.07^{a}$	0.09 <sup>a</sup>	$0.08^{a,b}$	$0.06^{b}$	$0.05^{b}$
Commonsono D	1470	3.18±	3.20±	3.24±	3.28±	3.29±	3.33±	3.39±
Germacrene D	14/9	0.14 <sup>a</sup>	$0.08^{a}$	0.09 <sup>a</sup>	0.11 <sup>a</sup>	0.10 <sup>a</sup>	0.09 <sup>a</sup>	$0.08^{a}$
& Cusions	1509	1.88±	1.89±	1.92±	1.94±	1.95±	1.99±	2.12±
<i>o</i> -Gualelle	1308	0.10 <sup>a</sup>	0.09 <sup>a</sup>	$0.07^{a}$	0.12 <sup>a</sup>	0.11 <sup>a</sup>	$0.08^{a}$	0.09 <sup>a</sup>
S Cadinana	1524	2.42±	2.45±	$2.49\pm$	2.50±	2.54±	2.59±	2.66±
o-Caumene	1324	0.10 <sup>a</sup>	$0.08^{a}$	$0.06^{a}$	0.09 <sup>a</sup>	$0.08^{a}$	$0.08^{a}$	$0.07^{a}$

RI: relative index (Kovats retention index); TIC: total ion current; nd – not determined; The results are presented as mean  $\pm$  SD (n=3)

a, b, c, d, e Values with different letters in a raw are statistically different (Tukey's HSD test, p < 0.05)

The major primary acids in wine grapes and subsequently in wines are tartaric, malic and depending from the grape variety but usually in minor amounts, citric acid (Bellman and Gallander, 1979). During the winemaking process and mainly fermentation, lactic, succinic, acetic and other acids could be formed and they play significant role in the final wine quality. The amounts of tentatively determined tartaric and malic acid are comparable for all variants of aromatized vines with control rosé with slight significant increase in W5 and W6. Malic acid is an important precursor of lactic acid through malolactic fermentation and the lactic acid formed is giving milder acidic taste (Bellman and Gallander, 1979). The amount of lactic acid found in all the variants is comparable which suggested that the added rose wastes of grape must did not influenced substantially the fermentation process.

A total of 41 volatile substances were tentatively detected (alcohols, acids, aldehydes, hydrocarbons and terpenes). Significant effect on the aroma substances formation/extraction in the aromatized wines and control were observed (Table 5) for  $\beta$ -citronellol, phenethyl alcohol, rose oxides, and geraniol. These substances were absent or present in the control in low amounts and appeared in the aromatized wines due to addition of rose waste. Phenethyl alcohol is among the compounds which contribute significantly to the favorable aroma of white and rosé wines (Cabaroglu and Canbas, 2002).  $\beta$ caryophyllene,  $\beta$ -citronellol, phenethyl alcohol, rose oxides, and geraniol increased significantly and distinctive rose aroma in W2-W6 variants was sensed. The amounts of alcohols increased in the aromatized wines compared to control but this increase was insignificant. The amount of acetaldehyde increased significantly for W4-W6 variants which confirm the GC-FID analysis results.

#### 3.2. Sensory analysis of aromatized wines

An important attribute of every new or modified food system is the consumers' opinion and for this reason in the subsequent experiments sensory analysis of aromatized wines was conducted (Figure 1). The results of the sensory tests revealed most of the panelists preferred W1 and W2 variants as wines with characteristics closer to the control rosé. This could be explained with the more traditionally oriented taste of the Bulgarian consumers concerning wines. The variants W3 to W6 were characterized with more pronounced rose aftertaste, the grassy nuances became more intense, the bitterness increased (along with the astringency, although astringency was not included in the indicators of the sensory analysis but most of the panelists expressed such sensations), as well as the flowery nuances increased. The W5 and W6 were disliked by most of the panelists (Overall acceptability  $2.1\pm0.8$  and  $2.0\pm0.8$ , respectively, compared with  $6.1\pm1.2$  for the control). In general the panelists divided wines in three groups: 1). Control, W1 and W2; 2). W3 and W4; and 3). W5 and W6. The group one was preferred mostly by the traditionally oriented consumers. It is interesting to note the opinion of some of the consumers towards group 3: they gave highest marks to these variants based on their personal preferences for aromatized (especially with rose notes) low-alcoholic beverages.



Figure 1. Sensory profile of wines

#### 4. Conclusions

The present work explored the possibility for preparation and managing the quality of aromatized wines with addition of essential rose oil industry waste during grape must fermentation. To the best of our knowledge this is the first attempt for manufacturing of aromatized wines with addition of essential rose oil industry waste and the first experimental data for GC-MS profiling of Mavrud rosé. Control rosé and six variants with addition of different amounts of dry rose wastes (from 0.05% to 2%) were obtained. The aromatized wines and the control had alcoholic content in the 14.4-14.6 $\pm$ 0.2 % range and pH 3.43 $\pm$ 3.59 $\pm$ 0.1. The polyphenol content increased significantly from 355.01 $\pm$ 10.14 to 576.08 $\pm$ 12.08 µmol GAE L<sup>-1</sup> for the control and W6, respectively. The higher amounts of phenolic acids (mainly gallic acid and 3,4-dihydroxy benzoic acid: from 1.2 $\pm$ 0.9 mg L<sup>-1</sup> and 5.4 $\pm$ 0.2 mg L<sup>-1</sup> in the control to 25.9 $\pm$ 0.9 and 65.1 $\pm$ 1.1 mg L<sup>-1</sup> in W6, respectively) is also related to increase in the antioxidant capacity of aromatized wines from

W1 to W6 and this could be explained with the higher amounts of total polyphenols extracted from the rose waste but not the anthocyanins (no significant difference in the TMA amounts). The aromatized wines and the control rosé had a total amounts of fusels in the 117.4-129.0±1.5 mg L<sup>-</sup> <sup>1</sup> range and it could be concluded that addition of rose waste during fermentation did not affect negatively formation of higher alcohols. βcaryophyllene, β-citronellol, phenethyl alcohol, rose oxides, and geraniol increased significantly and rose aroma in W1-W6 was achieved. The sensory analysis revealed W1 and W2 (overall acceptability  $6.6\pm1.0$  and  $5.9\pm0.9$ , respectively) were considered with closer characteristics to control wine (overall acceptability  $6.1\pm1.2$ ) and more appropriate for consumption by the consumers. The overall interpretation of experimental data suggested that added rose wastes in the grape must during its fermentation, did not influenced substantially the fermentation process. The results of the present study confirmed the main hypothesis that rose oil industry by-products successfully could be utilized for preparation of aromatized wine and contributed for augmentation of total polyphenol content, antioxidant capacity, and new aroma profile of the final product was obtained.

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